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DICTIONARY FILE UPDATES: 4 MAR 2008 HIGHEST RN 1006657-22-2

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=> d que stat 15

G1 \$



CH3 811

VAR G1=H/11
NODE ATTRIBUTES:
CONNECT IS EZ RC AT 4
DEFAULT MLEVEL IS ATOM
GGCAT IS SAT AT 4
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 11

STEREO ATTRIBUTES: NONE L5 62762 SEA FILE=REGIS

L5 62762 SEA FILE=REGISTRY SSS FUL L3

100.0% PROCESSED 394644 ITERATIONS SEARCH TIME: 00.00.04 62762 ANSWERS

=> d que stat 110 L10 ST

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GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 30

STEREO ATTRIBUTES: NONE

=> d que stat 115 L15 STR



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GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 30

STEREO ATTRIBUTES: NONE

=> d que stat 123 L23 STR

10/588,210

3 Deleted: 3/6/2008

VAR G1=H/CH3 NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 14

STEREO ATTRIBUTES: NONE

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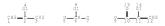
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STEREO ATTRIBUTES: NONE

L5 62762 SER FILE=REGISTRY SSS FUL L3

L12 SCR 2026 OR 1313

L24 STR



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GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 14

STEREO ATTRIBUTES: NONE

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100.0% PROCESSED 945 ITERATIONS SEARCH TIME: 00.00.01 200 ANSWERS

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FILE COVERS 1907 - 6 Mar 2008 VOL 148 ISS 10 FILE LAST UPDATED: 5 Mar 2008 (20080305/ED)

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=> d his

(FILE 'HOME' ENTERED AT 09:49:59 ON 06 MAR 2008)

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E US20070135560/PN 1 S E3

SEL RN

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FILE 'LREGISTRY' ENTERED AT 10:08:44 ON 06 MAR 2008

FILE 'REGISTRY' ENTERED AT 10:21:15 ON 06 MAR 2008

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SAV L5 TEMP RED210/A

1 S L2 AND L5

FILE 'LREGISTRY' ENTERED AT 10:37:56 ON 06 MAR 2008

L8 STR

FILE 'REGISTRY' ENTERED AT 11:05:40 ON 06 MAR 2008

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L10 STR L7 L11 46 S L10 SSS SAM SUB=L5 L12 SCR 2026 OR 1313

3 S L10 NOT L12 SSS SAM SUB=L5

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3/11/2008
                                 10/588.210
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L26
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L27
L28
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                SAV L28 RED210S4/A
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              39 S (L14 OR L17) AND ETHOXY
              57 S (L26 OR L28) AND ETHOXY
              57 S L29 OR L31
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              26 S L33 OR L34
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L36
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L35 ANSWER 1 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN
                         2007:935013 HCAPLUS <u>Full-text</u>
ACCESSION NUMBER:
DOCUMENT NUMBER:
                          147:285289
                          Eugenol-derived monomers and acrylic polymers
                          and compositions for biomedical uses
INVENTOR(S):
                          Rojo Del Olmo, Luis; Vazquez Lasa, Maria Blanca;
San Roman Del Barrio, Julio; Deb, Sanjukta
PATENT ASSIGNEE (S):
                          Consejo Superior de Investigaciones Cientificas,
                          Spain
SOURCE:
                          PCT Int. Appl., 50pp.
                          CODEN: PIXXD2
DOCUMENT TYPE:
                          Patent
LANGUAGE:
                          Spanish
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
     PATENT NO.
                          KIND
                                 DATE
                                              APPLICATION NO.
                                                                      DATE
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WO 2007093662

A1

20070823

WO 2007-ES70031

200702

5 \_\_\_ Deleted: 3/6/2008

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N: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BB, BH, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, BE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GT, HN, HR, HU, ID, IL, IM, IS, JP, RE, KG, KM, KN, PK, RR, KZ, IA, IC, LK, IR, IS, LT, LU, IV, IV, MA, MD, MB, KK, NN, MM, MX, MY, MZ, NA, NG, NI, NO, NZ, CM, PG, PH, PL, PT, SO, BS, HU, SC, SD, SE, SC, SK, SI, SM, SV, SY, JU, TM, IN, NR, TT, TZ, UA, UG, US, UZ, VC, VN, ZA, ZM, ZM, IS, ST, ST, TL, TL, II, UA, LV, CM, PR, EF, FI, FR, GB, CR, KT, TL, TL, UB, LV, NL, PL, PT, SO, SS, SI, SK, KT, IR, BS, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, CR, KT, IR, BS, CH, CY, CZ, DC, NL, PL, PT, SO, SS, SI, SK, KT, IR, BS, CH, CP, CG, CI, CX, CA, CN, CG, CM, ML, MS, NE, SN, TD, IS, SH, CT, SH, CH, CM, KE, LS, MI, MZ, MR, SI, SI, SZ, TZ, UG, ZM, AR, AZ, SY, NG, KZ, MD, RU, TJ, TN
```

AB The eugenol-derivative monomers have vinyl and methacrylate groups and alkyl substituents, e.g., eugenyl methacrylate, ethosy-eugenyl methacrylate and are prepared by reaction of eugenol or alkyl derivative with methacrylaty chloride. The eugenyl methacrylate or derivative and Me methacrylate or several methacrylate or derivative polymers comprise the eugenyl methacrylate or Several M

(eugenyl methacrylate monomers and copolymers and self-curable compns. for dental and bone reconstruction)

RN 912479-75-5 HCAPLUS

3/11/2008

CN 2-Propenoic acid, 2-methyl-, 2-[2-methoxy-4-(2-propen-1vl)phenoxylethyl ester, homopolymer (CA INDEX NAME)

CM 1 CRN 912479-73-3 CMF C16 H20 O4



II 913479-77-72, Ethoxy-eugenyl methacrylate-ethyl methacrylate copolymer

RL: IMF (Industrial manufacture); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(eugenyl methacrylate monomers and copolymers and self-curable

compns. for dental and bone reconstruction) RN 912479-77-7 HCAPLUS 7 Deleted: 3/6/2008

2-Propenoic acid, 2-methyl-, ethyl ester, polymer with 2-[2-methoxy-4-(2-propen-1-yl)phenoxy]ethyl 2-methyl-2-propenoate (CA INDEX NAME)

CM CRN 912479-73-3 CMF C16 H20 O4

CM

CRN 97-63-2 CMF C6 H10 O2

H2C 0

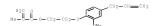
912479-73-3P, Ethoxy-eugenyl methacrylate RL: IMF (Industrial manufacture); RCT (Reactant); THU (Therapeutic

use); BIOL (Biological study); PREP (Preparation); RACT (Reactant or reagent); USES (Uses) (monomer; eugenyl methacrylate monomers and copolymers and

self-curable compns. for dental and bone reconstruction)

912479-73-3 HCAPLUS

2-Propencic acid, 2-methyl-, 2-[2-methoxy-4-(2-propen-1yl)phenoxy]ethyl ester (CA INDEX NAME)



63-7 (Pharmaceuticals)

Section cross-reference(s): 35

912479-74-4P, Eugenyl methacrylate homopolymer 913479-75-5P , Ethoxy-eugenyl methacrylate homopolymer

RL: IMF (Industrial manufacture); PREP (Preparation)

(eugenyl methacrylate monomers and copolymers and self-curable compns, for dental and bone reconstruction)

912479-76-6P, Ethyl methacrylate-eugenyl methacrylate copolymer 912474-77-7F, Ethoxy-eugenyl methacrylate-ethyl methacrylate copolymer

RL: IMF (Industrial manufacture); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(eugenyl methacrylate monomers and copolymers and self-curable compns. for dental and bone reconstruction)

T 375856-97-6P, Eugenyl methacrylate 915479-73-3P, Ethoxy-eugenyl methacrylate

RL: IMF (Industrial manufacture); RCT (Reactant); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(monomer; eugenyl methacrylate monomers and copolymers and self-curable compns, for dental and bone reconstruction)

REFERENCE COUNT: 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD, ALL CITATIONS AVAILABLE IN THE REFORMAT

L35 ANSWER 2 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2006:1079275 HCAPLUS Full-text

DOCUMENT NUMBER: 146:62955

TITLE: Solution- and Solid-Phase, Modular Approaches for Obtaining Different Natural Product-Like

Polycyclic Architectures from an Aminoindoline Scaffold for Combinatorial Chemistry

AUTHOR(S): Reddy, P. Thirupathi; Quevillon, S.; Gan, Zhonghong; Forbes, Nauzer; Leek, Donald M.;

Arya, Prabhat

CORPORATE SOURCE: Steadie Institute for Molecular Sciences,

National Research Council of Canada, Ottawa, ON, K1A OR6, Can.

SOURCE: Journal of Combinatorial Chemistry (2006), 8(6), 856-871

CODEN: JCCHFF; ISSN: 1520-4766
PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal
LANGUAGE: English
OTHER SOURCE(S): CASREACT 146:62955

GI

- AB With the goal of developing a modular approach leading to different indoline alkaloid natural-product-like tricyclic derives, having an unsatd. lactam, an aminoindoline-based bicyclic scaffold I was obtained from aminoindole II. The selective deprotection of the indoline N-Teoc or benzylio NiAlloc in compound I, followed by N-acryloylation and then subjection to a ring-closing metathesis reaction, successfully led to obtaining two different architectures having an unsatd. lactam functionality, e.g. III. This modular solution-phase methodol, was then developed on solid phase. To achieve this objective, the aminoindoline bicyclic scaffold having an addnl. hydroxyl group could be immobilized onto the solid support using alkylailyl linker-based polystyrene macrobeads. By applying a ring-closing metathesis approach, a tricyclic derivative with seven-membered-ring unsatd. lactam and a tricyclic derivative with seven-membered-ring unsatd. lactam metather a tricyclic derivative with seven-membered-ring unsatd. lactam were then obtained from in a number of steps.
- IT 910658-23-6P 916658-38-3F

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(solution and solid-phase preparation of natural product-like polycyclic compds. from aminoindole scaffold for combinatorial chemical)

RN 916658-23-6 HCAPLUS

<sup>\*</sup> STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY - AVAILABLE VIA OFFLINE PRINT \*

3/11/2(8)\$ 10/588,210 9 — Deleted 3/6/2008

CN Carbamic acid, N-[(2S,3S)-2-[(1R)-1-(acetyloxy)-3-buten-1-yl]-2,3-dihydro-5-[(2-methoxyethoxy)mthoxy]-1-[1-oxo-2-propen-1-yl]-1H-indol-3-yl]-, 2-propen-1-yl ester (CA INDEX NAME)

Absolute stereochemistry.

RN 916658-38-3 HCAPLUS

CN Carbamic acid, N=[(28,38)=2-[(18)-1-(acetyloxy)-3-buten-1-y1]-2,3-dihydro-5-[(2-methoxyethoxy)methoxy]-1-(1-oxo-2-propen-1-y1)-1H-indol-3-y1]-, 2-propen-1-y1 ester (CA INDEX NAME)

Absolute stereochemistry.

CC 31-5 (Alkaloids)

850559-86-3P 916658-22-5P 916458-13-68 916658-26-9P 916658-28-1DP, resin-bound 916658-28-1P 916658-29-2DP, 916658-30-5DP, resin-bound 916658-31-6DP. resin-bound 916658-32-7DP, resin-bound resin-bound 916658-35-0P 916658-37-2P 916658-33-3P 916658-39-4P 916658-40-7P 916658-41-8P 916658-42-9P 916658-43-0P 916658-44-1P 916658-46-3DP, resin-bound 916658-45-2P 916658-47-4DP. resin-bound 916658-48-5DP, resin-bound 916658-49-6DP, resin-bound 916658-50-9DP resin-bound 916658-51-0DP, resin-bound 916658-52-1DP, resin-bound

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(solution and solid-phase preparation of natural product-like polycyclic compds. from aminoindole scaffold for combinatorial chemical)
REFERENCE COUNT: 28 THERE ARE 28 CITED REFERENCES AVAILABLE

THERE ARE 28 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L35 ANSWER 3 OF 26 HCAPLUS COPYRIGHT 2008 ACS ON STN ACCESSION NUMBER: 2006:935643 HCAPLUS Full-text

DOCUMENT NUMBER: 145:425614

TITLE: From Natural Products to Polymeric Derivatives of "Eugenol": A New Approach for Preparation of

Dental Composites and Orthopedic Bone Cements

Rojo, Luis; Vazquez, Blanca; Parra, Juan; Lopez Bravo, Antonio; Deb, Sanjukta; San Roman, Julio CORPORATE SOURCE: Institute of Polymer Science and Technology, CSIC, Madrid, 28006, Spain

SOURCE: Biomacromolecules (2006), 7(10), 2751-2761 CODEN: BOMAF6; ISSN: 1525-7797

American Chemical Society

DOCUMENT TYPE: LANGUAGE:

English Polymers with eugenol moieties covalently bonded to the macromol, chains were synthesized for potential application in orthopedic and dental cements. First, eugenol was functionalized with polymerizable groups. The synthetic methods employed afforded two different methacrylic derivs., where the acrylic and eugenol moieties were either directly bonded, eugenyl methacrylate (EgMA), or separated through an oxyethylene group, ethoxyeugenyl methacrylate (EEgMA). A typical Fisher esterification reaction was used for the synthesis of EgMA and EEgMA, affording the desired monomers in 80% yields. Polymerization of each of the novel monomers, at low conversion, provided soluble polymers consisting of hydrocarbon macromols, with pendant eugenol moieties. At high conversions only cross-linked polymers were obtained, attributed to participation of the allylic double bonds in the polymerization reaction. In addition, copolymers of each eugenol derivative with Et methacrylate (EMA) were prepared at low conversion, with the copolymn. reaction studied by assuming the terminal model and the reactivity ratios determined according to linear and nonlinear methods. The values obtained were rEgMA = 1.48, rEMA = 0.55 and rEEGMA = 1.22, rEMA = 0.42. High mol. weight polymers and copolymers were obtained at low conversion. Anal, of thermal properties revealed a Tg of 95 °C for PEgMA and of 20 °C for PEEgMA and an increase in the thermal stability for the eugenol derivs. polymers and copolymers with respect to that of PEMA. Water sorption of the copolymers was found to decrease with the eugenol derivative content. Both monomers EgMA and EEgMA showed antibacterial activity against Streptococcus mutans, producing inhibition halos of 7 and 21 mm, resp. Finally, cell culture studies revealed that the copolymers did not leach any toxic eluants and showed good cellular proliferation with respect to PEMA. This study thus indicates that the eugenvl methacrylate derivs, are

potentially good candidates for dental and orthopedic cements. RL: BSU (Biological study, unclassified); PRP (Properties); SPN

(Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses) (preparation of dental composites and orthopedic bone cements from

polymeric derivs. of the natural product eugenol) RN 912479-75-5 HCAPLUS

2-Propenoic acid, 2-methyl-, 2-[2-methoxy-4-(2-propen-1yl)phenoxy]ethyl ester, homopolymer (CA INDEX NAME)

CM

CRN 912479-73-3

CMF C16 H20 O4

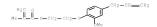
3/11/2008 10/588,210 11 -{ Deleted: 3/6/2008

RN 912479-77-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, ethyl ester, polymer with 2-[2-methoxy-4-(2-propen-1-yl)phenoxy]ethyl 2-methyl-2-propenoate (CA INDEX NAME)

CM

CRN 912479-73-3 CMF C16 H20 O4



м :

CRN 97-63-2 CMF C6 H10 O2



TT 912429-73-38

RL: BSU (Biological study, unclassified); RCT (Reactant); SPN (Synthetic preparation); BIOL (Biological study); PREP

(Preparation); RACT (Reactant or reagent)

(preparation of dental composites and orthopedic bone cements from polymeric derivs. of the natural product eugenol)

RN 912479-73-3 HCAPLUS

CN 2-Propencic acid, 2-methyl-, 2-[2-methoxy-4-(2-propen-1-yl)phenoxy]ethyl ester (CA INDEX NAME)



CC 63-7 (Pharmaceuticals)

Section cross-reference(s): 25, 35

912479-74-4P 912479-75-5P 912479-76-6P

912479-77-79

RL: BSU (Biological study, unclassified); PRP (Properties); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(preparation of dental composites and orthopedic bone cements from polymeric derivs. of the natural product eugenol)

polymeric derivs. of the natural product eugenol) II 375856-97-6P 912479-73-3P

RL: BSU (Biological study, unclassified); RCT (Reactant); SPN

3/13/2005 10/588,210 12 Deleted 3/6/2008

(Synthetic preparation); BIOL (Biological study); PREP

(Preparation); RACT (Reactant or reagent)

(preparation of dental composites and orthopedic bone cements from

polymeric derivs. of the natural product eugenol)
REFERENCE COUNT: 35 THERE ARE 35 CITED REFERENCES AVAILABLE

FOR THIS RECORD. ALL CITATIONS AVAILABLE
IN THE RE FORMIT

L35 ANSWER 4 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2006:826263 HCAPLUS Full-text

DOCUMENT NUMBER: 145:438789

TITLE: Part 2: Building Diverse Natural-Product-Like
Architectures from a Tetrahydroaminoquinoline

Scaffold. Modular Solution- and Solid-Phase Approaches for Use in High-Throughput Generation

of Chemical Probes

AUTHOR(S): Sharma, Utpal; Srivastava, Stuti; Prakesch, Michael; Sharma, Maya; Leek, Donald M.; Arya,

Prabhat
CORPORATE SOURCE: Steacie Institute for Molecular Sciences,

National Research Council of Canada, Ottawa, ON,

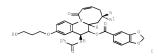
K1A OR6, Can.

SOURCE: Journal of Combinatorial Chemistry (2006), 8(5), 735-761

CODEN: JCCHFF; ISSN: 1520-4766

PUBLISHER: American Chemical Society
DOCUMENT TYPE: Journal

LANGUAGE: English
OTHER SOURCE(S): CASREACT 145:438789



AB The solution—and solid-phase synthesis to obtain several natural-productlike, tetrahydroquinoline-based, polycyclic derivs. were developed. In one approach, two derivs. I (R = OAc, Rl = H; R = H, Rl = OAc), having an eightmembered unsatd. lactam, were successfully obtained both in solution and on solid support.

IT 912628-35-6P 915603-59-2P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(solution- and solid-phase synthesis of tetrahydroquinoline-based natural-product-like polycyclic derivs.)

RN 912628-55-8 HCAPLUS

CN Oxazolo[5,4-c]quinoline-1(2H)-carboxylic acid, 4-[(2S)-2-(acetyloxy)-4-pentenyl]-3a,4,5,9b-tetrahydro-8-[(2-methoxyethoxy)methoxy]-2,2-

3/11/2008 10/588,210

13 \_\_\_ Deleted: 3/6/2008

dimethyl-5-(1-oxo-2-propenyl)-, phenylmethyl ester, (3aS,4S,9bS)-(9CI) (CA INDEX NAME)

Absolute stereochemistry.

RN 912628-59-2 HCAPLUS

CN Oxazolof5, 4-dquinoline-1(2H)-carboxylic acid, 4-[(2H)-2-(acetyloxy)-4-pentenyl]-3a, 45, 9b-tetrabydro-8-[(2-nethoxyethoxy)]-x2-dimethyl-5-[1-oxo-2-propenyl)-, phenylmethyl ester, (3a8, 48, 9b8)-(9CI) (CA INDEX NAME)

Absolute stereochemistry.

31-5 (Alkaloids) 912483-62-6P 912483-77-3P 912628-45-6P 912628-46-7P 912628-47-8P 912628-50-3P 912628-52-5P 912628-54-7P 912628-58-1P 312628-59-2P 912628-61-6DP, resin-bound 912628-61-6P 912628-62-7DP. resin-bound 912628-63-8DP, resin-bound 912628-65-0DP, resin-bound 912628-65-0P 912628-66-1DP, resin-bound 912628-67-2DP, resin-bound 912628-70-7DP, resin-bound 912628-73-0P 912628-70-7P 912628-72-9P 912628-74-1P 912628-75-2P 912628-76-3P 912628-77-4P 912628-78-5P 912628-79-6P 912628-80-9P 912628-81-0P 912628-82-1P 912628-83-2P 912628-84-3P 912628-85-4P 912628-86-5P 912628-87-6P 912628-88-7P 912628-89-8P 912628-90-1P 912628-91-2P 912628-92-3P 912628-93-4P 912628-94-5P 912628-96-7P 912628-97-8P 912628-98-9P 912628-95-6P 912628-99-0P 912629-00-6DP, resin-bound 912629-01-7DP, resin-bound 912629-02-8DP, resin-bound 912629-03-9DP,

3/11/2008 10/588.210 14 Deleted: 3/6/2008

resin-bound 912629-04-0DP, resin-bound 912629-05-1DP, 912629-12-0DP, resin-bound 912629-22-2DP, resin-bound resin-bound 912629-24-4DP, resin-bound resin-bound 912629-25-5DP, resin-bound

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(solution- and solid-phase synthesis of tetrahydroquinoline-based natural-product-like polycyclic derivs.)

REFERENCE COUNT: 25 THERE ARE 25 CITED REFERENCES AVAILABLE

FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L35 ANSWER 5 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2006:269793 HCAPLUS Full-text

DOCUMENT NUMBER: 144:340792

Lithography technique using silicone molds INVENTOR(S): Bahadur, Maneesh; Chen, Wei; Albaugh, John;

Harkness, Brian; Tonge, James

PATENT ASSIGNEE (S): Dow Corning Corporation, USA SOURCE: PCT Int. Appl., 28 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent English LANGUAGE:

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.				KIN	D	DATE APPLICATION NO.				D	ATE					
	2006		55		A2		2006	0323		WO 2	005-	US31	150		2	00508
WO	2006	0314	55		A.3		2006	1026								
		CH, GB, KP, MN, SC, UG, AT, IE, BF, TG,	CN, GD, KR, MX, SD, US, BE, IS, BU,	CO, GE, KZ, MZ, SE, UZ, BG, IT, CF, GH,	CR, GH, LC, NA, SG, VC, CH, LT, CG, GM,	CU GM LK NG SK VN CY LU CI KE	AU, CZ, HR, LR, NI, SL, YU, CZ, LV, CM, LS,	DE, HU, LS, NO, SM, ZA, DE, MC, GA, MW,	DK, ID, LT, NZ, SY, ZM, DK, NL, GN,	DM, IL, LU, OM, TJ, ZW EE, PL, GQ, NA,	DZ, IN, LV, PG, TM, ES, PT, GW, SD,	EC, IS, MA, PH, TN, FI, RO, ML,	EE, JP, MD, PL, TR, FR, SE, MR,	EG, KE, MG, PT, TT, GB, SI, NE,	ES, KG, MK, RO, TZ, GR, SK, SN,	FI, KM, MN, RU, UA, HU, TR, TD,
DD.	1803			AZ,	BY,		KZ,					7034	0.2			
LI	1003	055			7.2		2007	0101		LL Z	005-	1333	02		3	00508 1
	R:						CZ,									
CN	1010		4		A		2007	0815		CN 2	005-	8003	0625		2	00508
US	2007	2697	47		A1		2007	1122		US 2	007-	6599	89			00702
KR	2007	0523	05		A		2007	0521		KR 2	007-	7058	58		2	00703

3/11/2008 10/588,3

10/588,210 15 \_\_\_\_ Deleted: 3/6/2008

US 2004-609425P

200409 13

WO 2005-US31150

200508

- AB A method includes the steps of: (A) filling a silicone mold having a patterned surface with a curable (meth)acrylate formulation, (B) curing the curable (meth)acrylate formulation to form a patterned feature, (C) separating the silicone mold and the patterned feature, optionally (D) etching the patterned feature, and optionally (E) repeating steps (A) to (D) reusing the silicone mold. The curable (meth)acrylate formulation contains a fluorefunctional
- (meth)acrylate, a (meth)acrylate, and a photoinitiator.

RL: NUU (Other use, unclassified); USES (Uses)
(lithog, technique using silicone molds)

RN 208995-35-1 HCAPLUS

PRIORITY APPLN. INFO.:

CN 2-Propencic acid, 2-methyl-, methyl-2-[methyl-2-(2-propenyloxy)ethoxy]ethyl ester (CA INDEX NAME)

2 (D1\_Me)

CC 74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

75-91-2D, 1,1-Dimethylethyl hydroperoxide, reaction product with bis[2,2,6,6-tetramethyl-1-(octyloxy)-4-piperidinyl] ester 78-08-0, Vinyltriethoxysilane 78-10-4, Tetraethoxysilane 79-10-7, Acrylic methacrylate 97-63-2. Ethyl methacrylate 97-86-9. Isobutyl methacrylate 97-88-1, n-Butyl methacrylate 97-90-5, Ethylene glycol dimethacrylate 97-99-4 101-43-9, Cyclohexyl methacrylate 105-16-8, Diethylaminoethyl methacrylate 106-63-8, Isobutyl acrylate 106-74-1, 2-Ethoxyethyl acrylate 106-91-2, Glycidyl methacrylate 107-98-2, 1-Methoxy-2-propanol 108-46-3, 1,3-Benzenediol, uses 109-16-0, Triethylene glycol dimethacrylate 109-17-1, Tetraethylene glycol dimethacrylate 111-20-6, Decanedicic acid, uses 112-53-8, 1-Dodecanol 119-53-9D, Benzoin, ether 119-61-9, Benzophenone, uses 126-98-7, Methacrylonitrile 142-90-5, Lauryl methacrylate 407-47-6 502-44-3D, Caprolactone, acrylate deriv 585-07-9, tert-Butyl methacrylate 611-73-4 681-84-5, Tetramethoxysilane 688-84-6, 2-Ethylhexyl methacrylate 818-61-1, 2-Hydroxyethyl acrylate 868-77-9, 2-Hydroxyethyl methacrylate 923-26-2, 2-Hydroxypropyl methacrylate 947-19-3, 1-Hydroxycyclohexyl phenyl ketone 989-38-8, Rhodamine 6G 999-61-1, 2-Hydroxypropyl acrylate 1070-70-8, 1,4-Butanediol 1330-61-6, Isodecyl acrylate 1663-39-4, diacrylate 1189-08-8 tert-Butyl acrylate 2082-81-7, Butanediol dimethacrylate 2156-97-0, Lauryl acrylate 2223-82-7, Neopentyl glycol diacrylate 2358-84-1. Diethylene glycol dimethacrylate 2370-63-0, 2-Ethoxyethyl methacrylate 2399-48-6, Tetrahydrofurfuryl acrylate

2439-35-2, Dimethyl aminoethyl acrylate 2455-24-5, Tetrahydrofurfuryl methacrylate 2461-18-9D, [(Dodecyloxy)methyl]oxirane, reaction product with 4-[4,6-bis(2,4-dimethylphenyl)-1,3,5-triazin-2-yl] 2478-10-6, 4-Hydroxybutyl acrylate 2495-25-2, Tridecyl methacrylate 2495-35-4, Benzyl acrylate 2495-37-6, Benzyl methacrylate 2530-83-8, Glycidoxypropyltrimethoxysilane 2530-85-0 2602-34-8 2768-02-7, Vinyltrimethoxysilane 2867-47-2, Dimethyl aminoethyl methacrylate 3066-71-5, Cyclohexyl acrylate 3076-04-8, Tridecyl acrylate 3121-61-7, 2-Methoxyethyl acrylate 3290-92-4,
Trimethylol propane trimethacrylate 3524-68-3, Pentacrythritol triacrylate 4074-88-8, Diethylene glycol diacrylate 4491-03-6, Bisphenol A diacrylate 4813-57-4, Stearyl acrylate 4986-89-4, Pentaerythritol tetraacrylate 5039-78-1 5888-33-5, Isobornyl acrylate 6606-59-3, 1,6-Hexanediol dimethacrylate 6652-28-4, Benzoin isopropyl ether 7328-17-8 7473-98-5. 2-Hydroxy-2-methyl-1-phenylpropan-1-one 7534-94-3, Isobornyl methacrylate 7779-31-9 9016-00-6D, Polydimethylsiloxane, polyether-modified 13048-33-4, 1,6-Hexanediol diacrylate 13402-02-3, Cetyl acrylate 15206-55-0, Methylbenzoyl formate 15625-89-5, Trimethylol propane triacrylate 15895-80-4 17831-71-9, Tetraethylene glycol diacrylate 19485-03-1. 1,3-Butylene glycol diacrylate 21142-29-0, 3-Methacryloxypropyltriethoxysilane 22499-12-3, Benzoin isobutyl ether 24615-84-7, 2-Carboxyethyl acrylate 24650-42-8, Benzildimethylketal 25154-39-6, Tetrafluoropropyl acrylate 25721-76-0, Polyethylene glycol dimethacrylate 25736-86-1 25736-86-1 25852-49-7, Polypropylene glycol dimethacrylate 26570-48-9, Polyethylene glycol diacrylate 27458-06-6, Benzovlbenzoic acid 27905-45-9 28961-43-5 29570-58-9D. Dipentaerythritol hexaacrylate, caprolactone modified 29590-42-9, Isooctyl acrylate 31621-69-9 31900-57-9, Polydimethylsiloxane 32171-39-4 32360-05-7, Stearyl methacrylate 36811-99-1, 2,2'-(2,5-Thiophendivl)bis(tert-butylbenzoxazole) 38056-88-1 38785-10-3, Trifluoroethyl methacrylate 39420-45-6, Polypropylene glycol monomethacrylate 39670-09-2 41637-38-1, Bisphenol A ethoxylate dimethacrylate 41680-37-9D, Dipentaerythritol hexamethacrylate, caprolactone modified 42594-17-2, Tricyclodecane dimethanol diacrylate 42978-66-5, Tripropylene glycol diacrylate 48145-04-6, 2-Phenoxyethyl acrylate 51728-26-8 52408-84-1 52408-84-1 53879-54-2 56093-53-9, Pentaerythritol acrylate 57472-68-1, Dipropylene glycol diacrylate 60506-81-2, Dipentaerythritol pentaacrylate 61253-00-7, Octafluoropentyl methacrylate 64111-89-3, Dipropylene glycol dimethacrylate 64401-02-1 67362-76-9, 2-Butoxyethyl-4-dimethylaminobenzoate 72829-09-5 73507-02-5, Methyl benzoylbenzoate 75577-70-7, Trimethylolpropane ethoxy triacrylate 82799-44-8 83846-85-9, 4-Benzoyl-4'-methyldiphenyl sulfide 84170-74-1 92933-79-4, Octafluoropentyl acrylate 94108-97-1, Ditrimethylolpropane tetraacrylate 119313-12-1 162881-26-7, Phenylbis(2,4,6-trimethylbenzoyl)phosphine oxide 208995-35-1 236422-51-8, Octyl decyl acrylate 880485-38-1 880485-39-2D, reaction product with [(dodecyloxy)methyl]oxirane RL: NUU (Other use, unclassified); USES (Uses) (lithog, technique using silicone molds)

L35 ANSWER 6 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2005;23389 HCAPLUS Fill-text
DCCUMENT NUMBER: 142:115027
ITILE: Production of organic glasses containing low

3/11/2008 10/588,210 17 -{ Deleted: 3/6/2008

amounts of residual monomers

Arzhakov, M. S.; Arzhakov, S. A.; D'yachkov, A. I.; D'yachkov, I. A.; Skorobogatova, A. E.;

Stovachenko, I. L.; Chernavin, V. A.

PATENT ASSIGNEE(S): Russia

SOURCE: Russ., No pp. given COPEN: RUXXE7

DOCUMENT TYPE: Patent
LANGUAGE: Russian
FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
RU 2243978	C2	20050110	RU 2001-121094	200107
PRIORITY APPLN. INFO.:			RU 2001-121094	30 200107

AB A polymer for manufacturing organic glasses is produced by radical polymerization of Me methacrylate or its mixts, with (meth)acrylates or vinyl acctate in the presence of a radical initiator to obtain a polymer-monomer mixture, followed by extrusion with simultaneous depolymen. The radical initiator is a mixture of at least two initiators with different decomposition temps, or an initiator with two different decomposition temps. The method provides polymers containing low ants, of residual monomers (monomer conversions close to 100%). Thus, Me methorylate was polymerized at 230° in monomer conversion of 90%, followed by extrusion at 110° with simultaneous decolumn, to a monomer conversion of 90%, followed by extrusion at 110° with simultaneous decolumn, to a monomer conversion of 90% and monomer of provided the provided by the provided provided by the

II 53935-94-7DP, Me methacrylate-based polymers 59935-96-9DP, Me methacrylate-based polymers

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(production of organic glasses containing low amts. of residual monomers)

RN 58985-94-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-(2-propenyloxy)ethoxy]ethyl ester (9CI) (CA INDEX NAME)

RN 58985-96-9 HCAPLUS

N 2-Propenoic acid, 2-methyl-, 2-[2-[2-(2-

propenyloxy)ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

H2 C O O O C H2 — C H2

IC ICM C08F265-06

ICS C08F002-46; C08J003-28; B29C071-00; B29C055-12

3/11/2008 18 - Deleted: 3/6/2008

37-3 (Plastics Manufacture and Processing)

80-62-6DP, Methyl methacrylate, polymers 96-05-9DP, Allyl methacrylate, Me methacrylate-based polymers isocyanurate, Me methacrylate-based polymers 2998-04-1DP, Diallyl adipate, Me methacrylate-based polymers 16839-48-8DP, Me methacrylate-based polymers 26330-22-3DP, Me methacrylate-based polymers 26872-73-1DP, Me methacrylate-based polymers

58985-94-70P, Me methacrylate-based polymers 18385-96-30F, Me methacrylate-based polymers RL: IMF (Industrial manufacture); TEM (Technical or engineered

material use); PREP (Preparation); USES (Uses) (production of organic glasses containing low amts, of residual monomers)

L35 ANSWER 7 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN 2004:1041465 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER:

Comb-shaped single ion conductors based on

polyacrylate ethers and lithium alkyl sulfonate AUTHOR(S): Sun, Xiao-Guang; Hou, Jun; Kerr, John B. CORPORATE SOURCE: EETD, MS 62-203, Lawrence Berkeley National Laboratory, Berkeley, CA, 94720, USA

Electrochimica Acta (2005), 50(5), 1139-1147 CODEN: ELCAAV; ISSN: 0013-4686

PUBLISHER: Elsevier B.V. DOCUMENT TYPE:

AB

RN

LANGUAGE: English Comb-shaped single ion conductors have been synthesized by (1) sulfonation of small mol, chloroethyleneglycols, which, after ion exchange to the Li+ salt were then converted to the acrylate by reaction with acryloyl chloride and copolymd. with polyethylene glycol monomethyl ether acrylate (Mn = 454, n = 8) (PAES-co-E3SO3Li); (2) sulfonation of chloride end groups grafted on to prepolymers of polyacrylate ethers (PAES-q-EnSO3Li, n = 2, 3). The highest conductivity at 25 °C of 2.0 + 10-7 S cm-1 was obtained for the PAE8-co-E3SO3Li with a salt concentration of EO/Li = 40. The conductivity of PAE8-g-E3SO3Li is lower than that of PAE8-co-E3SO3Li at similar salt concns., which is related to the incomplete sulfonation of the grafted polymer that leads to a lower concentration of Li+. The addition of 50 weight% of plasticizer, PC/EMC (1/1, volume/volume), to PAE8-q-E2SO3Li increases the ambient conductivity by three orders of magnitude, which is due to the increased ion mobility in a micro-liquid environment and an increase concentration of free ions as a result of the higher dielec. constant of the solvent. A sym. Li/Li cell with an electrolyte membrane consisting of 75 weight % PC/EMC (1/1, volume/volume) was cycled at a c.d. of 100 mA cm-2 at 85 °C. The cycling profile showed no concentration polarization after a break-in period during the first few cycles, which was apparently due to reaction of the solvent at the lithium metal surface that reacted with lithium metal to form a stable SEI

laver. 535628-76-7DP, Diethylene glycol allyl ether

acrylate-oxirane graft copolymer, methyl ether, reaction products with {3-[2-(2-chloroethoxy)ethoxy]propy1}-1,1,3,3tetramethyldisiloxane and sodium sulfonate, ion-exchanged, lithium

RL: PRP (Properties); SPN (Synthetic preparation); PREP

(Preparation) (comb-shaped single ion conductors based on polyacrylate ethers and lithium alkyl sulfonate)

835628-76-7 HCAPLUS

2-Propenoic acid, 2-[2-(2-propenyloxy)ethoxy]ethyl ester, polymer with oxirane, methyl ether, graft (9CI) (CA INDEX NAME)

19 - Deleted: 3/6/2008

CM

CRN 67-56-1 CMF C H4 O

CM

CRN 835628-75-6

CMF (C10 H16 O4 . C2 H4 O)x CCI PMS

CM

CRN 286834-16-0 CMF C10 H16 O4

H2C - CH-CH2-O-CH2-CH2-O-CH2-CH2-O-C-CH-CH2

CM

CRN 75-21-8 CMF C2 H4 O

## $^{\circ}$

37-3 (Plastics Manufacture and Processing) Section cross-reference(s): 52

835628-37-OP 835628-38-1DP, reaction products with allyl ether-containing polyacrylate and sodium sulfonate, ion-exchanged, lithium salts 835628-39-2DP, reaction products with (3-[2-(2-chloroethoxy)ethoxy]propyl}-1,1,3,3-tetramethyldisiloxane and sodium sulfonate, ion-exchanged, lithium salts 835628-74-5P, Ethylene oxide-lithium 2-[2-(2-acryloylethoxy)ethoxy]ethyl sulfonate graft copolymer, methyl ether 835426-76-7DP, Diethylene glycol allyl ether acrylate-oxirane graft copolymer, methyl ether, reaction products with {3-[2-(2-chloroethoxy)ethoxy]propyl}-1,1,3,3tetramethyldisiloxane and sodium sulfonate, ion-exchanged, lithium salts RL: PRP (Properties); SPN (Synthetic preparation); PREP

(Preparation)

(comb-shaped single ion conductors based on polyacrylate ethers and lithium alkyl sulfonate)

REFERENCE COUNT: THERE ARE 35 CITED REFERENCES AVAILABLE FOR THIS RECORD, ALL CITATIONS AVAILABLE 3/11/2008 20 - Deleted: 3/6/2008 10/588.210

IN THE RE FORMAT

L35 ANSWER 8 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2004:249875 HCAPLUS Full-text DOCUMENT NUMBER: 140+272482

Aqueous (meth)acrylate compositions for water-resistant products with good colorant dispersion stability

INVENTOR(S): Fukada, Akihiko; Awaji, Toshio; Yoshimune, Soki PATENT ASSIGNEE(S): Nippon Shokubai Co., Ltd., Japan

Jpn. Kokai Tokkyo Koho, 20 pp.

CODEN: JKXXAF

DOCUMENT TYPE: LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2004091757	A	20040325	JP 2002-258847	200209
JP 4052904 PRIORITY APPIN. INFO.:	В2	20080227	JP 2002-258847	0.4
PRIORIII APPEN. INFO.:			OF 2002-230047	200209

Title compns., useful for jet-printing inks, coatings, adhesives, resists, etc., contain acetal- and/or hemiacetal ester bond-containing (meth) acrylates and optionally photopolymn, initiators. Thus, a reaction product of 2vinyloxyethoxyethyl methacrylate with tetraethylene glycol was mixed with water, Irgacure 3050 (photopolymn. initiator), and Benzopurpurin 4B (red direct dve) and left for 30 min to show no separation of the dve. The composition was applied on copying paper and UV cured to give a waterresistant coating with no discoloration after 1 min in water.

39385-94-70P, reaction products with polyols or carboxy-containing compds. 286854-16-80P, reaction products

with polyols or carboxy-containing compds.

RL: IMF (Industrial manufacture); RCT (Reactant); TEM (Technical or engineered material use); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(aqueous compns. containing acetal and/or hemiacetal ester bond-containing (meth)acrylates for water-resistant inks with good colorant dispersion stability)

RN 58985-94-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-(2-propenyloxy)ethoxy]ethyl ester (9CI) (CA INDEX NAME)

286834-16-0 HCAPLUS

2-Propenoic acid, 2-[2-(2-propenyloxy)ethoxy]ethyl ester (9CI) (CA CN INDEX NAME)

H2C CH CH2-O-CH2-CH2-O-CH2-CH2-O-U-CH-CH-CH2

IC ICM C08F299-00 ICS C08F020-28

CC 42-12 (Coatings, Inks, and Related Products)

Section cross-reference(s): 37

50-70-4DP, D-Sorbitol, reaction products with vinyl ether bond-containing acrylates 112-60-7DP, Tetraethylene glycol, reaction products with vinyl ether bond-containing (meth)acrylates 765-12-8DP, Triethylene glycol divinyl ether, reaction products with triethylene glycol mono (meth) acrylate 818-61-1DP, 2-Hydroxyethyl acrylate, reaction products with triethylene glycol divinyl ether and tetraethylene glycol 9004-75-5DP, Polyethylene glycol monosuccinate, reaction products with vinvl ether bond-containing acrylates 19812-60-3DP, Tetraethylene glycol monoacrylate, reaction products with triethylene glycol unsatd. ethers 21217-75-4DP, Tetraethylene glycol monomethacrylate, reaction products with triethylene glycol unsatd. ethers 25618-55-7DP, Polyglycerine, reaction products with vinvl ether bond-containing acrylates 50586-59-9DP, Polyethylene glycol trimethylolpropane ether, reaction products with vinyl ether bond-containing acrylates 39381-94-70P, reaction products with polyols or carboxy-containing compds. 76392-22-8DP, reaction products with polyols or carboxy-containing compds. 86273-46-3DP, reaction products with polyols or carboxy-containing compds. 90736-68-8DP. 4,7,10,13-Tetraoxahexadeca-1,15-diene, reaction products with triethylene glycol mono(meth)acrylate 123831-04-9P 286834-16-009, reaction products with polyols or carboxy-containing compds. 673477-34-4DP, reaction products with vinyl ether bond-containing acrylates RL: IMF (Industrial manufacture); RCT (Reactant); TEM (Technical or engineered material use); PREP (Preparation); RACT (Reactant or reagent); USES (Uses)

(aqueous compns. containing acetal and/or hemiacetal ester bond-containing (meth)acrylates for water-resistant inks with good colorant dispersion stability.

L35 ANSWER 9 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2004:158181 HCAPLUS Full-text

DOCUMENT NUMBER: 140:357767
TITLE: Synthesis and characterization of network type

single ion conductors
AUTHOR(S): Sun, Kiao-Guang; Reeder, Craig L.; Kerr, John B.

CORPORATE SOURCE: Lawrence Berkeley National Laboratory, Berkeley,

CA, 94720, USA
SOURCE: Macromolecules (2004), 37(6), 2219-2227

CODEN: MAMOBX; ISSN: 0024-9297
PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

LANSUNGE: English

8 New single ion conductors were synthesized by grafting the allyl groupcontaining lithium salt, lithium bis(allylmalonato)borate (LiBAMS), onto allyl
group-containing comb-branch polyacrylate or polymethacylate ethers by means
of hydrositylation. The highest ambient temperature conductivity of 3.5 \* 107.5 cm-l was obtained for a polyacrylate ether-based single ion conductivity.

containing eight EO units in the side chain and five EO units in the crosslinking side chain, to which the anion was fixed with a salt

concentration of Eo/Li = 20. For polyacrylate ether-based single ion conductors, an increase of chain length in both side chains and crosslinking anion chains favors an increase of ionic conductivity. The addition of 50 weight % EO/CMC (1/1, weight/weight) increased the lonic conductivity by more than 2 orders of magnitude due to both the increase in ionic mobility from the liquid phase and the increase in the concentration of free ions from the high dielec. constant of the solvent. The preliminary 11/Li cycling profiles of dry polyacrylate— and polymethacrylate ether-based single ion conductors are constant of the solvent of the solvent of the constant of the solvent of the solvent of the constant of the constant of the solvent of the preliminary 11/Li cycling profiles of the polymethacrylate ether-based single ion conductors are constant of the solvent of the so

IT 3:

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(monomer; for preparation of polyacrylate-polymethacrylate-based network-type single ion conductors)

RN 681819-03-4 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

H2C \_\_\_ CH\_\_ CH2\_ O\_\_ CH2\_\_ CH2\_\_ O\_\_ CH2\_\_ CH2\_\_ O\_\_ CH2\_\_ CH2\_\_ O\_\_ CH\_\_\_ CH\_\_\_\_ CH\_\_\_ CH\_\_\_\_ CH\_\_\_ CH\_\_\_\_ CH\_\_\_ CH\_\_\_\_ CH\_\_\_ CH\_\_

TT \$91919404-7P 681919-10-7

RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(synthesis and characterization of polyacrylate-polymethacrylatebased network-type single ion conductors)

681819-06-7 HCAPLUS

CN 2-Propenoic acid, 2-[2-(2-methoxyethoxy)ethoxy]ethyl ester, polymer with 2-(2-propenyloxy)ethyl 2-propenoate (9CI) (CA INDEX NAME)

CM

CRN 48067-72-7

CMF C10 H18 O5

MeO-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-CH

CM 2

CRN 7784-80-7 CMF C8 H12 O3

----

H2C-CH-CH2-O-CH2-CH2-O-CH-CH2-CH2

681819-10-3 HCAPLUS

2-Propencic acid, 3,6,9,12,15,18,21,24-octaoxapentacos-1-yl ester, polymer with 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl 2-propenoate (9CI) (CA INDEX NAME)

CRN 681819-08-9

CMF C20 H38 O10

Me O - CH2 - CH2 - O - CH2 - C

PAGE 1-B

- CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-

CM

CRN 681819-03-4

CMF C12 H20 O5

H2C - CH2 -

35-4 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 52

7784-80-7P, 2-Allyloxyethyl acrylate 16839-48-8P, 2-Allyloxyethyl methacrylate 48067-72-7P 51382-35-5P 681313-93-4P 681819-04-5P RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);

RACT (Reactant or reagent) (monomer; for preparation of polyacrylate-polymethacrylate-based

network-type single ion conductors) 681819-09-0P

681819-04-7P 681819-07-8P 581819-10-5P 681819-11-4P 681819-12-5P

RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(synthesis and characterization of polyacrylate-polymethacrylatebased network-type single ion conductors)

REFERENCE COUNT: 30 THERE ARE 30 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE

IN THE RE FORMAT

L35 ANSWER 10 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2001:842320 HCAPLUS Full-text

DOCUMENT NUMBER: 135:376826 3/11/2008 10/588,210 24 -{ Deleted: 3/6/2008

TITLE: Siloxanyl-containing monomers, their polymers, and ophthalmic lenses from the polymers

INVENTOR(S): Nakamura, Masataka; Yokota, Mitauru

PATENT ASSIGNEE(S): Toray Industries, Inc., Japan

SOURCE: Upn. Kokai Tokkyo Koho, 10 pp.

SOURCE: Jpn. Kokai Tol
CODEN: JKXXAF
DOCUMENT TYPE: Patent
LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2001323024	A	20011120	JP 2000-140441	200005
PRIORITY APPLN. INFO.:			JP 2000-140441	12
				200005

OTHER SOURCE(S): MARPAT 135:376826

AB XO(CHIZCHRO)m (CHIZ)nA (X = group having polymerizable C-C unsatd. bond; R = H, Me; A = alloxany; m, n = 2-10), polymers containing the monomers, and ophthalmic lenses made from the polymers are claimed. The lenses such as contact lenses have high O permeability, high water content, and low modulus of elasticity. Contact lenses were manufactured from CHIZ(MCACC) CRIZ/RIZ(OZ (CHIZ)RIZ)(CHIME) 3 (preparation given), N, N-

dimethylacrylamide, and triethylene glycol dimethacrylate. 55395.94-75

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation of siloxanyl-containing monomers, their polymers, and contact lenses therefrom)

RN 58985-94-7 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-(2-propenyloxy)ethoxy]ethyl ester (9CI) (CA INDEX NAME)

Me\_U\_U\_C\_O\_CH2\_CH2\_O\_CH2\_CH2\_O\_CH2\_CH

IC ICM C08F030-08

ICS A61L027-00; C07F007-08; C08F299-08; G02B001-04; G02C007-04

CC 63-7 (Pharmaceuticals)

Section cross-reference(s): 38

IT 15075-50-0P, Diethylene glycol monoallyl ether 58985-94-7P 374534-72-2P 374534-73-3P 374534-74-4P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);

RACT (Reactant or reagent)

(preparation of siloxanyl-containing monomers, their polymers, and contact lenses therefrom)

L35 ANSWER 11 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 2001:780791 HCAPLUS Full-text

DOCUMENT NUMBER: 135:319624

TITLE: Compositions for three-dimensional printing of solid objects

25 \_\_\_ Deleted: 3/6/2008 3/11/2008 10/588.210

INVENTOR(S): Bredt, James F.; Clark, Sarah L.; Uy, Evert F.; Dicologero, Matthew J.; Anderson, Timothy; Tarkenian, Michael

English

PATENT ASSIGNEE(S): Z Corporation, USA PCT Int. Appl., 34 pp. CODEN: PIXXD2 DOCUMENT TYPE: Patent

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

LANGUAGE:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2001078969	A2	20011025	WO 2001-US12220	200104
W0 2001078969 W: CA, JP, KR				
NL, PT, SE,	TR		FI, FR, GB, GR, IE, II	, LU, MC,
			CA 2001-2405539	200104 13
US 2001050031	A1	20011213	US 2001-835292	200104 13
EP 1272334	A2	20030108	EP 2001-927008	200104
	LT, LV	, FI, RO,	GB, GR, IT, LI, LU, NI MK, CY, AL, TR JP 2001-576254	, SE, MC,
				200104 13
US 2005197431	A1	20050908	US 2005-68487	200502 28
PRIORITY APPLN. INFO.:			US 2000-197118P	200004 14
			US 2000-197526P	P 200004 14
			US 2001-835292	B3 200104 13
			WO 2001-US12220	W 200104

AB The composition for selectively adhering particulate material to form a solid object in a three-dimensional printer comprises a nonaq. organic compound 108399-15-1, Di(propylene glycol) allyl ether methacrylate RL: TEM (Technical or engineered material use); USES (Uses)

<sup>(</sup>compns. for three-dimensional printing of solid objects) 208995-35-1 HCAPLUS

RN

3/11/2108 26 Deleted: 3/6/2008

CN 2-Propenoic acid, 2-methyl-, methyl-2-[methyl-2-(2-propenyloxy)ethoxy]ethyl ester (CA INDEX NAME)

 $\begin{array}{c} \text{H2C} \\ \text{Me} \\ \text{C} \\ \text{C}$ 

2 (D1-Me)

IC ICM B29C067-00 ICS B29C041-00; B41J002-01; C09D011-10

42-11 (Coatings, Inks, and Related Products) 64-17-5, Ethanol, uses 67-63-0, Isopropanol, uses 75-65-0, tert-Butanol, uses 97-88-1, Butyl methacrylate 106-65-0, Dimethyl succinate 111-55-7, Ethylene glycol diacetate 123-25-1, Diethyl succinate 141-78-6, Ethyl acetate, uses 471-34-1, Calcium carbonate, uses 544-17-2, Calcium formate 627-93-0. Dimethyl adipate 1189-08-8, 1,3-Butylene glycol dimethacrylate 1305-62-0, Calcium hydroxide, uses 1314-13-2, Zinc oxide, uses 1317-61-9, Iron oxide (Fe304), uses 1327-44-2, Potassium aluminum silicate 1330-43-4, Sodium tetraborate 1335-30-4, Aluminum silicate 1344-09-8, Sodium silicate 1344-28-1, Aluminum oxide, uses 1344-95-2, Calcium silicate 1985-51-9, Neopentyl glycol dimethacrylate 6484-52-2, Ammonium nitrate, uses 6606-59-3, 1,6-Hexanediol dimethacrylate 7558-79-4 7631-86-9, Silica, uses 7647-14-5, Sodium chloride, uses 7778-80-5, Potassium sulfate, uses 7783-28-0, Ammonium hydrogen phosphate 9002-89-5, Polyvinyl alcohol 9002-98-6 9003-20-7, Polyvinyl acetate 9003-39-8, Polyvinyl pyrrolidone 9003-53-6, Polystyrene 9003-56-9, ABS 9011-14-7, Polymethyl methacrylate 9080-79-9, Sodium polystyrene sulfonate 10042-91-8 11104-48-6, Calcium aluminate 12125-02-9, Ammonium chloride, uses 13048-33-4, 1,6-Hexanediol diacrylate 13463-67-7, Titania, uses 18023-33-1, Vinyltriisopropoxysilane 25086-89-9, Vinyl acetate-vinyl pyrrolidone copolymer 25087-2 Polymethacrylic acid 26062-79-3, Polydiallyldimethylammonium chloride 26124-23-2, Vinvlpvrrolidone-acrylamide copolymer 31113-94-7, Vinyl methyl ether-vinyl pyrrolidone copolymer 42978-66-5, Tri(propylene glycol) diacrylate 48145-04-6, Ethylene glycol phenyl ether acrylate 54193-36-1, Polymethacrylic acid sodium salt 208905-35-1, Di(propylene glycol) allyl ether methacrylate 367277-91-6, Vinylpyrrolidone-2-ethyl-2-oxazoline

RL: TEM (Technical or engineered material use); USES (Uses) (compns. for three-dimensional printing of solid objects)

L35 ANSWER 12 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1998:600088 HCAPLUS Full-text

DOCUMENT NUMBER: 129:291215

TITLE: Curable acrylic polysiloxane compositions for

INVENTOR(S): flexible weather-resistant coatings
174, Nobuc; Cosuqi, Koji
PATENT ASSIGNEE(S): Nippon Paint Co., Ltd., Japan
SOURCE: Jun. Kokai Tokkyo Koho. 7 pp.

Jpn. Kokai Tokkyo Koho, 7 pp. CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT:

27 Deleted: 3/6/2008

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 10245488	A	19980914	JP 1997-65402	
				199703 03
ORITY APPLN. INFO.:			JP 1997-65402	199703
				0.3

- The compns. comprise (a) polysiloxanes bearing ≥2 SiH groups/mol., (b) alkenyl group-containing acrylic polymers (iodine value 40-100, Mn 1000-100,000), and (c) hydrosilylation catalysts, where the mol. ratio of SiH in (a) to alkenyl in (b) is 0.5-4. Thus, a varnish of 2-allyloxyethyl methacrylate-styrenemethacrylic acid-cyclohexyl methacrylate copolymer (Mn 4100, iodine value 60) was mixed with di-Ph Me H polysiloxane, 2% EtoH solution of chloroplatinic acid, and 2-methyl-3-butyn-2-ol, applied on an Fe sheet, and baked to give clear coatings showing excellent acid resistance, pencil hardness H, and good resistance to xylene rubbing test.
- 214131-80-90F, reaction products with hydrogen polysiloxanes 214135-81-970, reaction products with hydrogen polysiloxanes 214193-31-329, reaction products with hydrogen polysiloxanes 216113-34-3DP, reaction products with hydrogen polysiloxanes RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (curable acrylic polysiloxane coatings with excellent weather resistance and flexibility)
- 214133-80-9 HCAPLUS RN

PATENT INFORMATION:

- 2-Propenoic acid, 2-methyl-, cyclohexyl ester, polymer with ethenylbenzene, methyl 2-methyl-2-propenoate and 2-[2-(2-propenyloxy)ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)
- CM
- CRN 58985-94-7 CMF C11 H18 O4
- H2C 0 Me—U—U—0—CH2—CH2—CH2—CH2—CH2—CH2—CH
  - CM
  - CRN 101-43-9 CMF C10 H16 O2

CM

CRN 100-42-5 CMF C8 H8

H2 C == CH=Ph

CM 4

CRN 80-62-6

CMF C5 H8 O2

Mc\_U\_U\_OMe

214133-81-0 HCAPLUS

214133-01-0 mcFMS 2-methyl-, cyclohexyl ester, polymer with ethenylbenzene, ethyl 2-propenoate, methyl 2-methyl-2-propenoate and 2-[2-(2-propenyloxy) ethyl (2-methyl-2-propend) (GA INDEX NAME)

CM

CRN 58985-94-7

CMF C11 H18 O4

CM 2

CRN 140-88-5 CMF C5 H8 O2

Eto\_C\_CH\_CH2

CM

CRN 101-43-9 CMF C10 H16 O2



CM

CRN 100-42-5 CMF C8 H8

H2 C --- CH-Ph

5 CM

CRN 80-62-6 CMF C5 H8 O2

H2C 0 0Me

214133-83-2 HCAPLUS

2-Propenoic acid, 2-methyl-, cyclohexyl ester, polymer with z-rropenois acid, z-methyl-, cyclonexyl ester, polymer wit ethenylbenzene, N-(1-methylethyl)-2-propenamide, methyl 2-methyl-2-propenoate and 2-[2-(2-propenyloxy)ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58985-94-7 CMF C11 H18 O4

 $\begin{array}{c} \text{H2C} & \text{O} \\ \text{Me} - \text{C} - \text{C} - \text{O} - \text{CH}_2 + \text{CH}_2 + \text{O} + \text{CH}_2 + \text{CH}_2 + \text{O} + \text{CH}_2 +$ 

CM 2

CRN 2210-25-5 CMF C6 H11 N O

T-PENH\_CH\_CH\_CH2

CM

CRN 101-43-9 CMF C10 H16 O2



CM

CRN 100-42-5 CMF C8 H8

H2C-CH-Ph

CM 5

CRN 80-62-6 CMF C5 H8 O2

Me\_U\_U\_OMe

214133-84-3 HCAPLUS

219133-94-3 MARIOS

-Propenoic acid, 2-methyl-, cyclohexyl ester, polymer with
ethenylbenzene, ethyl 2-propenoate, N-(1-methylathyl)-2-propenamide,
methyl 2-methyl-2-propenoate and 2-[2-(2-propenyloxy)ethoxylethyl
2-methyl-2-propenoate (9C1) (CA INDEX NAME)

CRN 58985-94-7

CMF C11 H18 O4

H2C Me-C-C-CH2-CH2-O-CH2-CH2-O-CH2-CH=-CH2-CH

CM 2

CRN 2210-25-5

CMF C6 H11 N O

1-PrNH\_C\_CH\_\_CH2

CM 3

CRN 140-88-5 CMF C5 H8 O2

Eto\_\_CH\_\_CH2

CM 4

CRN 101-43-9 CMF C10 H16 O2

CM 5

CRN 100-42-5 CMF C8 H8

CM 6

CRN 80-62-6

CMF C5 H8 O2

IC ICM C08L083-05

ICS C08G081-02; C08L029-10; C08L033-06; C08L033-14; C09D129-10;

3/11/2008 10/588.210 32 - Deleted: 9/6/2008

C09D133-06; C09D133-14; C09D183-05

42-7 (Coatings, Inks, and Related Products) 155904-19-1DP, Diphenylsilanediol-methylsilanediol copolymer, trimethylsilyl-terminated, reaction products with alkenyl-containing acrylic polymers 214133-64-9DP, 2-Allyloxyethyl methacrylate-cyclohexyl methacrylate-methyl methacrylate-styrene copolymer, reaction products with hydrogen polysiloxanes 214133-65-0DP, reaction products with hydrogen polysiloxanes 214133-66-1DP, reaction products with hydrogen polysiloxanes 214133-67-2DP, reaction products with hydrogen polysiloxanes 214133-69-4DP, 2-Allyloxyethyl acrylate-cyclohexyl methacrylate-methyl methacrylate-styrene copolymer, reaction products with hydrogen polysiloxanes products with hydrogen polysiloxanes products with hydrogen polysiloxanes products with hydrogen polysiloxane products with hydrogen polysiloxane 214133-71-8DP, reaction products with di-Ph Me H polysiloxane 214133-74-DDP, Cyclohexyl methacrylate-9-decenyl methacrylate-methyl methacrylate-styrene copolymer, reaction products with hydrogen polysiloxanes 214133-75-2DP, reaction products with hydrogen polysiloxanes 214133-76-3DP, reaction products with hydrogen polysiloxanes 214133-79-6DP, reaction products with hydrogen polysiloxanes 234133-80-95P, reaction products with hydrogen polysiloxanes 214133-41-35F, reaction products with hydrogen polysiloxanes 214133-85-200, reaction products with hydrogen polysiloxanes 214133-84-5DE, reaction products with hydrogen polysiloxanes

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (curable acrylic polysiloxane coatings with excellent weather resistance and flexibility)

L35 ANSWER 13 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1998:397820 HCAPLUS Full-text

DOCUMENT NUMBER: 129:68138

TITLE: Preparation of alkoxylated glycidyl (meth)acrylates

INVENTOR(S): Fan, Mingxin; Ceska, Gary W.; Horgan, James;

PATENT ASSIGNEE(S): Sartomer Co., USA
SOURCE: U.S., 4 pp.
CODEN: USXXAM
DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PA:	TENT :	NO.		KIN	D	DATE			APPL	ICAT	ION I	NO.		D.	ATE
					-										
	5763			A		1998	0609		US 1	996-	7729	79		1	99612
WO	9828	287		A1		1998	0702		WO 1	997-	EP72	33		1 2	99712
			BE,			PL, ES,		FR,	GB,	GR,	IE,	IT,	LU,	MC,	NL,
EP	9465		-	A1		1999	1006		EP 1	997-	9544	58		1 2	99 <b>7</b> 12

3/11/2008 10/588,210 33 Deleted: 3/6/2008

R: BE, DE, ES, FR, GB, IT, NL PRIORITY APPLN. INFO.:

US 1996-772979 A 199612

WO 1997-EP7283

W 199712 22

OTHER SOURCE(S): MARPAT 129:68138

AB Alkoxylated glycldyl (meth)acrylates are prepared by epoxidizing alkoxylated allyl (meth)acrylates with 1202 in the presence of (a) tungatic acid or its metal salts, (b) phosphoric sold or its metal salts, (b) phosphoric sold or its metal salts, and (c) &l phase transfer catalyst. Thus, 100.0 g propoxylated allyl methacrylate prepared from propoxylated allyl alc. and methacrylic acid, was mixed with 2.7 g MeM [(CI27)7513 Pok/Me(O(22)214 in 100 g toluene, and 50 ml. H202 (304) was added into the mixture in 30 min at 60° and reacted for 22.0 h to yield a epoxide at conversion 85%.

RN 208995-35-1 HCAPLUS CN 2-Propendic acid. 2-methyl-, methyl-2-[methyl-2-(2propenyloxy)ethoxy]ethyl ester (CA INDEX NAME)

2 (D1\_Me)

IC ICM C07D301-12 INCL 549531000

CC 35-2 (Chemistry of Synthetic High Polymers)

I 208995-34-0P 208905-35-1P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(Preparation); MACT (Reactant or reagent)
(preparation of alkoxylated glycidyl (meth)acrylates)
REFERENCE COUNT: 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR

THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L35 ANSWER 14 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1995:737303 HCAPLUS Full-text DOCUMENT NUMBER: 123:113219

TITLE: Silicon-modified acrylic copolymer and adhesive INVENTOR(S): Hosoi, Yasuhiro; Iwamoto, Osamu; Himeno,

Masataka
PATENT ASSIGNEE(S): Tokuyama Corp., Japan

SOURCE: Eur. Pat. Appl., 64 pp.

DOCUMENT TYPE: Patent
LANGUAGE: English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

	PATENT NO.	KIND	DATE	APPLICATION NO.		DATE
	PATENT NO.	KIND	DAIL	APPLICATION NO.		DATE
	EP 632060	A2	19950104	EP 1994-304825		400407
						199407
	EP 632060	A3	19950125			0.1
	EP 632060	B1	19981014			
	R: DE, FR, GB JP 07070246	A	10050214	JP 1994-67362		
	OF 07070Z40	P.	19930314	OF 1334-0130Z		199404
						0.5
	JP 3105733		20001106			
	JP 07076611	A	19950320	JP 1994-91902		199404
						28
	JP 3107702	В2	20001113			
	US 5476912	A	19951219	US 1994-268321		
						199406 30
PRIO	RITY APPLN. INFO.:			JP 1993-164640	Α	00
						199307
						0.2
				JP 1993-175176	A	
						199307
						15
				JP 1994-67362	A	
				DF 1394-01302	71	199404
						0.5
				JP 1994-91902		
				OP 1994-91902	A	199404
						28

AB A silicone-modified acrylic copolymer having a weight-average mol. weight (Mw) 5000-1,000,000, comprising (a) acrylate structural unit having an CG13-alkyl group or an aryl group having 6-14 C atoms; (b) a siloxane modified acrylate unit where siloxane is covalently bound through A which is a divalent organic group having 2-20 C atoms which may optionally comprise an ether bond or an ester bond in the main chain; optionally (c) acrylate structural unit having an ethylenically unsatd. hydrocarbon group of 2-20 C atoms which may optionally comprise an ether bond or an ester in the main chain with abice (10-99.9):(90-0.1);(0-89.9). An adhesive was prepared from the graft polymer (Mw 160,000) of Me3Sio(Me3Sio)(MeHSio)(10) MeHSio(10) MeB 27.9, catalyst 0.33, and ally methacrylate-Me methacrylate copolymer 5 g, dissolved in CHCIC2, applied on acrylic plate, and bonded with a silicone paste, showing adhesion (to acrylic plate even after 3 min warm water soaking) >20 Kg/cm2.

II 1585-98-909, graft polymer with SlH group-containing siloxane RL: IMF (Industrial manufacture); PRP (Properties); IEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (preparation and adhesion properties of)

RN 58985-96-9 HCAPLUS CN 2-Propenoic acid, 2-methyl-, 2-[2-[2-(2-

propenyloxy)ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

H2C 0

IC ICM C08F008-42 ICS C08G081-02

CC 35-8 (Chemistry of Synthetic High Polymers) Section cross-reference(s): 38, 63

Section Großa-Teierbendeij, 13, 03

18 8-62-69P, Methyl methacrylate, graft polymer with SiH group-containing alloxane 96-03-90P, Allyl methacrylate, graft polymer with SiH group-containing alloxane 96-03-90P, Allyl methacrylate, graft polymer with SiH group-containing siloxane 99-03-90P, Methyl acrylate, graft polymer with SiH group-containing siloxane 29-25-30P, Allyl acrylate, graft polymer with SiH group-containing siloxane 249-25-20P, graft polymer with SiH group-containing siloxane 249-25-20P, graft polymer with SiH group-containing siloxane 1533-09-90P, graft polymer with SiH group-containing alloxane 1533-09-90P, graft polymer with SiH group-containing alloxane 1993-90P, graft polymer with SiH group-containing alloxane 110083-27-70P, graft polymer with SiH group-containing siloxane 110083-27-70P, graft polymer with SiH group-containing siloxane 110083-27-70P, graft polymer with SiH group-containing siloxane RH: IMF (industrial namufacture), PREP (Properation); DMR (Chenical or engineered material use); PREP (Properation); USES (Uses) (preparation and abbeion properties of)

L35 ANSWER 15 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1991:196399 HCAPLUS Full-text

DOCUMENT NUMBER: 114:196399
TITLE: Waterless pr

TLE: Waterless presensitized lithographic plate with photosensitive layer containing allyloxyethyl actylate copolymer

INVENTOR(S): Azuma, Tatsuji; Kawamura, Koichi; Kita, Nobuyuki PATENT ASSIGNEE(S): Fuji Photo Film Co., Itd., Japan SOURCE: Jun. Kokai Tokkvo Koho, 11 pp.

SOURCE: Jpn. Kokai Tokkyo Koho, CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 02226248	A	19900907	JP 1989-46955	
				198902 28
RIORITY APPLN. INFO.:			JP 1989-46955	
				198902

28

IT 153411-75-6, 2-Allyloxyethyl methacrylate-methacrylic acid-triethyleneglycol monomethacrylate copolymer sodium salt RL: USES (Uses)

(lithog. plate photosensitive layer using)

N 133411-78-6 HCAPLUS

2-Propenoic acid, 2-methyl-, polymer with 2-(2-(2hydroxyethoxy)ethoxylethyl2-methyl-2-propenoate and 2-(2-propenyloxy)ethyl 2-methyl-2-propenoate, sodium salt (9CI) (CA INDEX NAME)

CM 1

CRN 133411-77-5

CMF (C10 H18 O5 , C9 H14 O3 , C4 H6 O2)x

CCI PMS

CM 2

CRN 16839-48-8 CMF C9 H14 O3

H2C 0 CH2 CH2 O CH2 CH2 CH2 CH2 CH2 CH

CM

CRN 2351-42-0

CMF C10 H18 O5

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CRN 79-41-4
CMF C4 H6 O2
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Ú\_co2H

ICM G03F007-00 ICS G03F007-038

CM

74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

26570-48-9 86829-42-7 133394-54-4, 2-Allyloxyethyl methacrylate-2-methacryloxyethylhydrogen succinate copolymer 133394-55-5, 2-Allyloxyethyl methacrylate-methacrylic acid copolymer 133394-56-6, 2-Acryloxyethylhydrogen succinate-2-allyloxyethyl methacrylate-2-hydroxyethyl methacrylate copolymer 13:471-78-5, 2-Allyloxyethyl methacrylate-methacrylic acid-triethyleneglycol monomethacrylate copolymer sodium salt 133536-76-2 RL: USES (Uses)

(lithog, plate photosensitive layer using)

L35 ANSWER 16 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN 1988:152251 HCAPLUS Full-text ACCESSION NUMBER: DOCUMENT NUMBER: 108:152251 TITLE: Urethane (meth) acrylates for coating materials Fukuchi, Shuzo, Yamaguchi, Shigeru INVENTOR(S): PATENT ASSIGNEE (S): Nippon Shokubai Kagaku Kogyo Co., Ltd., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 15 pp. CODEN: JKXXAF DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE JP 62230762 19871009 A. JP 1986-70883 198603 PRIORITY APPLN. INFO.: JP 1986-70883 198603

Urethane (meth) acrylates are prepared by reacting hydroxyalkyl (meth) acrylates with (un) substituted alkenyl glycidyl ethers and optionally other cyclic compds. and treating with other hydroxy compds. and polyisocyanates. Thus, 2hydroxyethyl acrylate was treated with allyl glycidyl ether at ratio 1:3 to give CH2:CHC02CH2CH2O [CH2CH(CH20CH2CH:CH2)0]3H, treated with 2,4-toluene diisocyanate to give a urethane acrylate, treated with 2-hydroxyethyl acrylate, mixed with pentaerythritol tetra(3-mercaptopropionate) and Irgacure 651, coated on steel, and irradiated with a high-pressure Hg lamp to form a coating having pencil hardness 5H.

RL: IMF (Industrial manufacture); RCT (Reactant); PREP

(Preparation); RACT (Reactant or reagent) (manufacture and reaction of, with polyisocyanates)

- 112861-63-9 HCAPLUS RN
- 2-Propenoic acid, 2-[2-[2-(2-hydroxymethylethoxy)methylethoxy][(2propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)
- HO-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O

H2C-CH-CH2-O-CH2-D1

2 (D1-Me)

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(manufacture and reaction of, with toluene diisocyanate) 112861-61-7 HCAPLUS

2-Propenoic acid, 2-methyl-, 2-[2-[2-[2-hydroxy](2-

propenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy][(2propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

3 H2C CH CH2 - O- CH2 - D1

- 112861-62-8 HCAPLUS
- 2-Propenoic acid, 2-[2-[2-[2-hydroxy[(2propenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy][(2-

propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

3 H2C=CH-CH2-O-CH2-D1

- ICM C07C125-06
- ICS C08G018-67
- ICA C08F299-06
- 42-10 (Coatings, Inks, and Related Products)
- Section cross-reference(s): 55

RL: IMF (Industrial manufacture); RCT (Reactant); PREP

3/11/2008 10/588,210 39 -{Deleted: 3/6/2008

(Preparation); RACT (Reactant or reagent)

(manufacture and reaction of, with polyisocyanates)

112861-61-7P 111801-62-8P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(manufacture and reaction of, with toluene diisocyanate)

L35 ANSWER 17 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1988:96319 HCAPLUS Full-text

DOCUMENT NUMBER: 108:96319

TITLE: Light-curable polyene-polythiol coating materials

INVENTOR(S): Fukuchi, Shuzo; Yamaguchi, Shigeru

PATENT ASSIGNEE (S): Nippon Shokubai Kagaku Kogyo Co., Ltd., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 10 pp.

CODEN: JKXXAF
DOCUMENT TYPE: Patent
LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 62241925	A	19871022	JP 1986-84328	198604
PRIORITY APPLN. INFO.:			JP 1986-84328	14

- - (Preparation); RACT (Reactant or reagent) (manufacture and reaction of, with toluene diisocyanate)
- RN 112861-62-8 HCAPLUS CN 2-Propencic acid, 2-[2-[2-hydroxy](2-
- cn Z-rropenoic acia, z-[z-[z-nyaroxy](zpropenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy][(2propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

IT 112861-61-7P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP

3/11/2008 Deleted: 3/6/2008 40

(Preparation); RACT (Reactant or reagent) (manufacture and reaction of, with toluene diisocyanate and

trimethylhexamethylene diisocyanate)

112861-61-7 HCAPLUS

2-Propenoic acid, 2-methyl-, 2-[2-[2-[2-hydroxy[(2propenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy][(2propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

H2C ĬŪ O-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-CH2-OH

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(manufacture and reaction of, with trimethylhexamethylene diisocyanate)

112861-63-9 HCAPLUS RN

2-Propenoic acid, 2-[2-[2-(2-hydroxymethylethoxy)methylethoxy][(2propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

HO-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-CH2-CH2-

## 2 (D1-Me)

ICM C08G075-04 42-10 (Coatings, Inks, and Related Products)

Section cross-reference(s): 55

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (manufacture and reaction of, with toluene diisocvanate)

RL: IMF (Industrial manufacture); RCT (Reactant); PREP

(Preparation); RACT (Reactant or reagent)

(manufacture and reaction of, with toluene diisocyanate and trimethylhexamethylene diisocyanate)

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (manufacture and reaction of, with trimethylhexamethylene

diisocyanate)

L35 ANSWER 18 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1988:96318 HCAPLUS Full-text

3/11/2008 41 Deleted: 3/6/2008 10/588.210

14

DOCUMENT NUMBER: 108:96318

Light curable polyene-polythiol coating materials

PATENT ASSIGNEE(S):

Fukuchi, Shuzo; Yamaquchi, Shiqeru Nippon Shokubai Kagaku Kogyo Co., Ltd., Japan

CODEN: JKXXAF DOCUMENT TYPE: Patent LANGUAGE: Japanese FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

SOURCE:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 62241924	A	19871022	JP 1986-84327	198604
PRIORITY APPLN. INFO.:			JP 1986-84327	14 198604

Jpn. Kokai Tokkyo Koho, 8 pp.



Coating materials contain compds. having ≥2 SH groups/mol. in alkenyl groupcontaining (meth)acrylates CH2:CRCO2ZOAlBmH (R = H or Me, Z = C2-8 divalent organic groups, A = ring-opened groups of alkenyl glycidyl ethers I, B = ringopened groups of cyclic compds. except I, 1 = 1-20, m = 0 or 1-20, the arrangement of A and B being arbitrary), and R1 = H or <C20 (halo substituted) hydrocarbyl groups and R2 <C20 alkenyl groups in I. The reaction of 2hydroxyethyl acrylate with allyl glycidyl ether with 98.1% CH2:CHCO2CH2CH2O[CH2CH(CH2OCH2CH:CH2)O]3H which (56 parts) was mixed with 44 parts pentaerythritol tetra(3-mercaptopropenoate), coated on steel, and irradiated with high-pressure Hg damp to form a coating having pencil hardness 3 H.

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(manufacture and photochem. polymerization of, with polydiols, for coatings)

RN 112861-61-7 HCAPLUS

CN

2-Propenoic acid, 2-methyl-, 2-[2-[2-hydroxy](2-propenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy[(2-propenyloxy)methyl propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

3/11/2008 10/588.210 42. Deleted: 3/6/2008

112861-62-8 HCAPLUS

2-Propenoic acid, 2-[2-[2-[2-hydroxy[(2propenyloxy)methyl]ethoxy][(2-propenyloxy)methyl]ethoxy][(2propenyloxy)methyllethoxylethyl ester (9CI) (CA INDEX NAME)

112861-63-9 HCAPLUS

2-Propenoic acid, 2-[2-[2-(2-hydroxymethylethoxy)methylethoxy][(2propenyloxy)methyl]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

HO-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-CH2

$${\rm H}_2{\rm C}$$
  ${\rm CH}_-{\rm CH}_2$   ${\rm O}$   ${\rm CH}_2$   ${\rm D}$   ${\rm D}$ 

## 2 (D1-Me)

ICM C08G075-04

42-10 (Coatings, Inks, and Related Products)

Section cross-reference(s): 55

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(manufacture and photochem. polymerization of, with polydiols, for coatings)

L35 ANSWER 19 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1986:412173 HCAPLUS Full-text

DOCUMENT NUMBER:

ORIGINAL REFERENCE NO.: 105:2033a,2036a

Curable resin compositions for dental use INVENTOR(S): Sakashita, Takeshi; Nakano, Takayuki Mitsui Petrochemical Industries, Ltd., Japan

PATENT ASSIGNEE(S):

SOURCE: Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent 3/11/2008

10/588.210

43 Deleted: 3/6/2008

LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 61036204	A	19860220	JP 1984-155621	
				198407
				27
JP 03078841	В	19911217		
PRIORITY APPLN. INFO.:			JP 1984-155621	
				198407
				2.7

AB Curable resin compns. for dental use comprise (1) poly(meth)acryloyloxyalkyl(iso)cyanurate I or II (R1-R3 = H, alkyl, acryloyl, methacryloyl; X1-X3 = oxyalkylene, polyoxyalkylene), (2) vinyl compds., and (3) a polymerization initiator. Thus, bis(methacryloyloxyethyl)(hydroxyethyl)isocyanurate (1 g), a vinyl compound (1 g), camphorquinone (6 mg), Me N, N-dimethylanthranilate (6 mg), and powdered silica (2 g) were mixed to form a paste, which was irradiated with visible light (350-700 nm) for 30 s. The Brinell hardness was 70 and the bending

strength was 1150 kg/cm2.

RL: BIOL (Biological study)

(dental composites or cements containing)

102770-35-4 HCAPLUS DN

2-Propenoic acid, 2-methyl-, 2-[2-[(3-methyl-3butenyl)oxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

ICM A61K006-08

ICA C08F220-40; C08F226-06

63-7 (Pharmaceuticals)

109-16-0 1565-94-2 3077-12-1 35838-12-1 56745-15-4 82508-13-2 102770-05-4

RL: BIOL (Biological study)

(dental composites or cements containing)

L35 ANSWER 20 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN 1985:542411 HCAPLUS Full-text ACCESSION NUMBER:

3/11/2008 10/588,210 44 Deleted: 3/6/2008

DOCUMENT NUMBER: 103:142411

ORIGINAL REFERENCE NO.: 103:22829a,22832a

TITLE: Synthesis of polymers containing crown lactone units via cyclopolymerization in the presence of

alkylaluminum chlorides
AUTHOR(S): Yokota, Kazuaki; Kakuchi, Tovoji; Taniguchi,

Yasuyuki, Takada, Yoshiyuki
CORPORATE SOURCE: Fac. Eng., Hokkaido Univ., Sapporo, 060, Japan
SOURCE: Makromolekulare Chemie, Rapid Communications

(1985), 6(3), 155-61 CODEN: MCRCD4; ISSN: 0173-2803

DOCUMENT TYPE: Journal LANGUAGE: English

AB Polymerization of the acrylates CH2:CHCH2(OCH2CH2)yO-o-

Folymerization of the acrylates CHRICHER/CORNECHE/DY-Co-GEHAC(HER/CHRICK)XCORCHER (x = 1-3, y = 1-2) or or CHRICHER/CER/CHRICK)XCORCHER (x = 2-5) by Al chloroalkyls gave cyclopolymers containing crown ether lactone units. Bt3Al2cl3 and EtAlCl2 were more effective than EtAlCl, but often gave insol. polymers. The polymns, were faster than radical cyclopolymns. In the extraction of alkali metal picrates the cation binding ability of the crown ether lactone derivative polymers decreased in the order: beno-21-crown beno-24-crown-7 beno-2-3-crown-6 > beno-14-crown-5 > beno-14-crown-4. For other crown ether polymers, stability was also

greatest for rings containing 6 0 atoms. IT 77504-04-2P 77504-06-9P 77504-08-4P

26387-13-6P 96387-35-8F 96187-37-0P

RL: SPN (Synthetic preparation); PREP (Preparation)
(crown ether lactone-containing, preparation and metal binding capacity of)

RN 77504-04-2 HCAPLUS

N 2-Propensic acid, 2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CRN 77504-03-1 CMF C16 H20 04

0-CH2-CH2-O-CH2-CH2-O-CH=-CH

CH2-CH\_CH2

RN 77504-06-4 HCAPLUS

CN 2-Propencic acid, 2-[2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM

CRN 77504-05-3

77504-08-6 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethoxy]et hoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CRN 77504-07-5

CMF C20 H28 O6

PAGE 1-A , 0— CH 2— CH 2— O— CH 2— CH 2— O— CH 2— C CH2-CH-CH2

PAGE 1-B

-CH2

96387-33-6 HCAPLUS

2-Propenoic acid, 2-[2-[2-(2-propenyloxy)ethoxy]phenoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 96387-32-5 CMF C16 H20 O5

96387-35-8 HCAPLUS

2-Propenoic acid, 2-[2-[2-[2-(2-propenyloxy)ethoxy]phenoxy]ethoxy]et hyl ester, homopolymer (9CI) (CA INDEX NAME)

CM

CRN 96387-34-7

3/11/2008

10/588.210

Deleted: 3/6/2008

CMF C18 H24 O6

96387-37-0 HCAPLUS RN

2-Propenoic acid, 2-[2-[2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]phenoxy ]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM

CRN 96387-36-9

CMF C20 H28 O7

RN 96387-39-2 HCAPLUS

2-Propencic acid, 2-[2-[2-[2-[2-[2-(2-propenyloxy)ethoxy]e

CM

CRN 96387-38-1

CMF C22 H32 O8

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of) 96387-32-5 HCAPLUS

2-Propenoic acid, 2-[2-[2-(2-propenyloxy)ethoxy]phenoxy]ethyl ester (9CI) (CA INDEX NAME)

96387-34-7 HCAPLUS

2-Propenoic acid, 2-[2-[2-[2-(2-propenyloxy)ethoxy]phenoxy]ethoxy]et CN hyl ester (9CI) (CA INDEX NAME)

RN96387-36-9 HCAPLUS

2-Propencic acid, 2-[2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]phenoxy ]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

96387-38-1 HCAPLUS RN

2-Propencic acid, 2-[2-[2-[2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]phen oxy]ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

- 35-3 (Chemistry of Synthetic High Polymers) CC
  - - 96387-39-3P 96387-41-6P
    - RL: SPN (Synthetic preparation); PREP (Preparation)
    - (crown ether lactone-containing, preparation and metal binding capacity
  - 26587-32-58196387-34-7E 96387-36-9P
  - - RL: SPN (Synthetic preparation); PREP (Preparation)
      - (preparation of)

3/13/2003 48 Deleted 3/6/2008

L35 ANSWER 21 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1984:438883 HCAPLUS Full-text

DOCUMENT NUMBER: 101:38883 ORIGINAL REFERENCE NO.: 101:6095a,6

TITLE: Studies on cyclopolymerization in the presence

of alkylaluminum chlorides. VII. Cyclopolymerizations of acrylates and

methacrylates containing oligooxyethylene units in the 11-20-membered-ring region

Yokota, Kazuaki; Kakuchi, Toyoji; Iiyama,

Takashi: Takada, Yoshiyuki

CORPORATE SOURCE: Fac. Eng., Hokkaido Univ., Sapporo, 060, Japan

Polymer Journal (Tokyo, Japan) (1984), 16(2),

CODEN: POLJB8; ISSN: 0032-3896

DOCUMENT TYPE: Journal LANGUAGE: English

ANDUAGE: English

An investigation of cyclopolymm. in the presence of alkylaluminum chlorides
was extended to 2-[2-(o-allylphenoxy)ethoxylethyl acrylate [77504-03-1] and
higher homologs containing oligooxyethylene units capable of forming 14-, 17and 20-membered rings, resp. Although the effect of alkylaluminum chlorides
was gradually reduced with increasing ring size, it was remarkable in the
formation of 14-membered rings for methacrylates and even the 20-membered
rings for acrylates. When combined with previous results for the analogs
containing oligomethylene units, the present data showed that oxyethylene
groups had favorable effects on cyclopolymm. A plot of the extent of
cyclization against ring size showed that monomers containing oligomethylene
units had spreater cyclization tendency than those containing oligomethylene
units had spreater cyclization tendency than those containing oligomethylene
units that or containing oligomethylene of the rate consist. For linear propagation and
propagation of the rate of the

11 to 20 for acrylates. IT 77504-03-1P 77504-05-3F 77504-07-5P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);

RACT (Reactant or reagent)
(preparation and properties and cyclopolymn. of)

RN 77504-03-1 HCAPLUS

CN 2-Propencic acid, 2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)



RN 77504-05-3 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethoxy]ethyl
ester (9CI) (CA INDEX NAME)

RN 77504-07-5 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-[2-[2-[2-[2-propenyl)phenoxy]etho

PAGE 1-A

O-CH2-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-CH

PAGE 1-B

-CH2

RN 90883-60-6 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-[2-[2-(2propenyl)phenoxy]ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

RN 90883-61-7 HCAPLUS

N 2-Propenoic acid, 2-methyl-, 2-[2-[2-[2-[2-(2propenyl)phenoxy]ethoxy]ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

O-CH2-CH2-O-CH2-CH2-O-CH

RN 90883-62-8 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-[2-[2-(2-

propenyl)phenoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

$$\begin{array}{c} \text{O-CH}_2-\text{CH}_2-\text{O-CH}_2-\text{CH}_2-\text{O-CH}_2-\text{O-CH}_2-\text{O-CH}_2\\ \text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{O-CH}_2-\text{O-CH}_2-\text{O-CH}_2-\text{O-CH}_2\\ \end{array}$$

CC 35-3 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 67

T 77504-03-1P 77504-05-3E 77504-07-5P 10833-60-6P 90883-6)-7P 90863-62-8P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and properties and cyclopolymn. of)

L35 ANSWER 22 OF 26 HCAPLUS COPYRIGHT 2008 ACS ON STN ACCESSION NUMBER: 1984:85273 HCAPLUS Full-text

DOCUMENT NUMBER: 100:85273 ORIGINAL REFERENCE NO.: 100:12917a,12920a

TITLE: Diethylene glycol methacrylate allyl ether.

INVENTOR(S): Voronina, T. A.; Fomina, N. V. PATENT ASSIGNEE(S): USSR

SOURCE: U.S.S.R. From: Otkrytiya, Izobret., Prom. Obraztay, Tovarnye Znaki 1983, (38), 85.

DOCUMENT TYPE: Patent
LANGUAGE: Russian

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

ATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
SU 1047898	A1	19831015	SU 1982-3435735	
				198205 11
PRIORITY APPLN. INFO.:			SU 1982-3435735	
				198205

- AB Title substance is prepared from diethylene glycol and allyl bromide by treating the resulting monoallyl ether of diethylene glycol with methacrylic chloride in the presence of a catalyst at 0-(-5)\*. The procedure is simplified and the yield is increased by condensing diethylene glycol with allyl bromide at 90-140° in the presence of metallic Cu in a polar solvent consisting of DNF or DNSO with subsequent cooling of the reaction material containing the monoallyl ether of diethylene glycol up to 0-(-5)\* and addition of methacrylic chloride.
  - 58985-94-78
  - RL: SPN (Synthetic preparation); PREP (Preparation)
  - (preparation of)
- RN 58985-94-7 HCAPLUS
- CN 2-Propenoic acid, 2-methyl-, 2-[2-(2-propenyloxy)ethoxy]ethyl ester
- (9CI) (CA INDEX NAME)

51 \_\_\_ Deleted: 3/6/2008

O- CH2-CH2-O-CH2-CH2-O-CH2-CH2-CH2

C07C069-587; C07C069-54

23-17 (Aliphatic Compounds)

RL: SPN (Synthetic preparation); PREP (Preparation)

L35 ANSWER 23 OF 26 HCAPLUS COPYRIGHT 2008 ACS on SIN 1981:175657 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER:

ORIGINAL REFERENCE NO.: 94:28725a,28728a

TITLE: Effect of alkylaluminum chlorides on the cyclopolymerization of unconjugated dienes

Yokota, Kazuaki; Kakuchi, Toyoji; Takada, Yoshiyuki

CORPORATE SOURCE: Fac. Eng., Hokkaido Univ., Sapporo, Japan SOURCE: Hokkaido Daigaku Kogakubu Kenkyu Hokoku (1980).

(102), 45-54 CODEN: HDKKAA; ISSN: 0385-602X

DOCUMENT TYPE:

LANGUAGE: Japanese



AB Unconjugated dienes such as  $\omega$ -(2-allylphenoxy)alkyl acrylates [o-CH2:CHCH2C6H40(CH2CH2) nO2CCH:CH2, n = 1,2,3,5] (I), and  $\omega$ -(2allylphenyl)oligooxyethylene acrylates [o-CH2:CHCH2C6H4(OCH2CH2)nO2CCH:CH2, n = 0,2,3,4] (II) were polymerized in the presence of AlEt2Cl [96-10-6], AlEt1.5C11.5 [12075-68-2], and AlEtC12 [563-43-9] to give polymers containing 7-20 membered rings [III, R = O(CH2CH2)n or (OCH2CH2)n]. The catalysts increased the reaction rate and the extent of cyclization in the case of I (n = 1,2,3) or II (n = 0,2,3) but for making larger rings, they were not effective. Copolymn. studies with 4-chlorostyrene [1073-67-2] suggested that the catalysts interact with both the double bonds in the same monomer

17504-03-1 77504-05-3 77504-07-5

RL: RCT (Reactant); RACT (Reactant or reagent) (cyclopolymn. of, in presence of ethylaluminum chlorides, mechanism of)

RN 77504-03-1 HCAPLUS

2-Propenoic acid, 2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

77504-05-3 HCAPLUS

CN 2-Propencic acid, 2-[2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethoxy]ethyl ester (9CI) (CA INDEX NAME)

77504-07-5 HCAPLUS

PAGE 1-A , 0— CH 2— CH 2— O— CH 2— CH 2— O— CH 2— CH 2— OH 2— CH 2— CH 2— O— ČH СН2—СН==СН2

PAGE 1-B

■ CH2

IT 77504-04-1P 77504-05-4P 77504-08-6P

RL: SPN (Synthetic preparation): PREP (Preparation) (preparation of)

77504-04-2 HCAPLUS

2-Propencic acid, 2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CRN 77504-03-1

CMF C16 H20 O4

RN 77504-06-4 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-[2-(2-propenyl)phenoxy]ethoxy]ethoxy]ethyl ester, homopolymer (9CI) (CA INDEX NAME)

CM :

CRN 77504-05-3

CMF C18 H24 O5

RN 77504-08-6 HCAPLUS

CN 2-Propenoic acid, 2-[2-[2-[2-[2-(2-propenyl)phenoxy]ethoxy]

CM 1

CRN 77504-07-5 CMF C20 H28 O6

PAGE 1-A

PAGE 1-B

■ CH 2

CC 35-4 (Synthetic High Polymers)

IT 14925-75-8 61632-59-5 61632-60-8 77504-03-1 77504-05-2 77504-07-5 77504-09-7 77505-38-5

RL: RCT (Reactant); RACT (Reactant or reagent)

(cyclopolymn. of, in presence of ethylaluminum chlorides, mechanism of)

IT 27101-85-5P 77504-04-2P 77504-06-4P

3/11/2008 10/588.210 54 Deleted: 3/6/2008

77504-10-0P 77504-11-1P 77504-12-2P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

L35 ANSWER 24 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1979:169022 HCAPLUS Full-text

DOCUMENT NUMBER: ORIGINAL REFERENCE NO.: 90:26851a,26854a

TITLE: Synthesis of esters of acrylic and methacrylic

acids containing allyl groups AUTHOR(S): Beshenova, E. P.; Etlis, V. S. USSR

CORPORATE SOURCE: Osnovnoi Organicheskii Sintez i Neftekhimiya

(1977), 8, 57-9 CODEN: OOSNDC: ISSN: 0321-2386

DOCUMENT TYPE:

LANGUAGE: Russian

The syntheses are described of 8 title compds. CH2:CRCO2ZCH2CH:CH2 [R = H, Me; AB Z = CH2CH2OCO2CH2CH2C, CH2CH(OH)CH2O, CH2CH(CH2OCH2CH:CH2)OCO2, CH(CH2OCH2CH:CH2)CH2O] from acrylic and methacrylic acids, their derivs., and derivs. of allyl alc.

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)

RN 2-Propenoic acid, 2-[[[2-(2-propenyloxy)ethoxy]carbonyl]oxy]ethyl

ester (9CI) (CA INDEX NAME)

RN 69936-67-0 HCAPLUS

2-Propenoic acid, 2-methyl-, 2-[[[2-(2-propenyloxy)ethoxy]carbonyl]o xy]ethyl ester (9CI) (CA INDEX NAME)

\_Й\_о\_сн2\_сн2\_о\_сн2\_сн<u>—</u>сн2

35-2 (Synthetic High Polymers)

Section cross-reference(s): 23 22214-16-0P 22214-17-1P 59936-66-9P 69936-67-0P 69936-68-1P 69936-69-2P RL: SPN (Synthetic preparation): PREP (Preparation)

(preparation of)

L35 ANSWER 25 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN 1976:151442 HCAPLUS Full-text ACCESSION NUMBER: DOCUMENT NUMBER: 84:151442

ORIGINAL REFERENCE NO.: 84:24621a.24624a

Crosslinked hydrogel copolymer material Howes, John G. B.; Da Costa, Nicholas M.; INVENTOR(S): Selway, Rupert A.; Potter, William D.

3/11/2008 SOURCE:

LANGUAGE:

10/588,210

Smith and Nephew Research Ltd., UK

55 Deleted: 3/6/2008

Ger. Offen., 17 pp. CODEN: GWXXBX

German

DOCUMENT TYPE: FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PATENT ASSIGNEE(S):

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 2529639	A1	19760122	DE 1975-2529639	197507
				03
DE 2529639	C2	19850725		
GB 1494641	A	19771207	GB 1974-29757	197407
ZA 7504036	A	19760526	ZA 1975-4036	0.4
				197506 24
US 4036814	A	19770719	US 1975-590037	24
				197506 25
TO 7502351	A	19760106	NO 1975-2351	197506
				30
NO 144888	В	19810824		
NO 144888	C	19811202		
DK 7503006	A	19760105	DK 1975-3006	197507
VL 7507914	A	19760106	NL 1975-7914	03
2 7307524	**	13700100	No 1919 1914	197507 03
R 2277110	A1.	19760130	FR 1975-20881	
				197507 03
FR <b>2277</b> 110	B1	19801003		
AU 7582722	A	19770106	AU 1975-82722	
				197507 03
CH 603708	A5	19780831	CH 1975-8688	
				197507 03
CS 194 <b>2</b> 31	B2	19791130	CS 1975-4749	
				197507 03
BE 831047	A1	19760105	BE 1975-158027	0.3
				197507
FI 7501961	A	19760105	FI 1975-1961	04
ET 1201361	n.	19760103	LI 19/3-1961	197507
	_			04
FI 59114 FI 59114	B	19810227 19810610		
SE 7507693	A	19760105	SE 1975-7693	
				197507 04
SE 407416	C	19790705		0.4

3/11/2008	1		10/588,210				
SE	407416	В	19790326				
JP	51030750	A	19760316	JP	1975-82013		197507
	58045689	В	19831012				04
BR	7504239	A	19760706	BR	1975-5429		197507
DD	123396	A5	19761212	DD	1975-187112		197507
CA	1037196	A1	19780822	CA	1975-230799		197507
PRIORITY	APPLN. INFO.:			GB	1974-29757	Α	197407 04
				GB	1975-17586	Α	197504

56 Deleted: 3/6/2008

AB Phenylethyl methacrylate (I), benzyl methacrylate, phenoxyethyl methacrylate, β-naphthyl methacrylate, or a similar methacrylate was copolymd. with virylpyrrolidone (II) and with allyl methacrylate (III), 3-allyloxy-2-hydroxypropyl methacrylate, 2-allyloxyethyl methacrylate, or a similar crosslinking monomer to prepare crosslinked hydrogel copylmers which absorbed 65-85% of a physiol. salt solution and were especially useful for the manufacture of contact lenses. Thus, a mixture of I 6.4, II 33.6, III 0.27, and AIBN 0.12 g were heated 24 hr at 45-55° and 1-10 hr at 110° to prepare a copolymer [58986-09-7].

COPOLYMET (5898-09-7).

IT 5845-94-5 58988-07-0 58985-98-1
59935-99-3 59986-00-8 58986-01-9
58986-02-7 59986-08-2 58986-05-3
38386-02-3 58986-08-6
RL: USES (Uses)

(hydrogels) RN 58985-95-8 HCAPLUS

National States of Stat

CM 1 CRN 58985-94-7 CMF C11 H18 04

CM 2

CRN 3683-12-3 CMF C12 H14 O2 H2C 0 4e-C-C-O-CH2-CH2-Ph

CM

CRN 88-12-0 CMF C6 H9 N O



RN 58985-97-0 HCAPLUS

NN 5995-9/-0 HOAFLOS CON 2-Propensic acid, 2-methyl-, 2-phenylethyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM

CRN 58985-96-9 CMF C13 H22 O5

CM 2

CRN 3683-12-3 CMF C12 H14 O2

H2C O CH2 CH2 Ph

CM

CRN 88-12-0 CMF C6 H9 N O



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58985-98-1 HCAPLUS
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2-Propencic acid, 2-methyl-, phenylmethyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2 propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM

CRN 58985-96-9 CMF C13 H22 O5

CM

CRN 2495-37-6

CMF C11 H12 O2

H2C 0 -0- CH2-Ph

CM

CRN 88-12-0 CMF C6 H9 N O

58985-99-2 HCAPLUS

2-Propenoic acid, 2-methyl-, phenyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2- propenyloxy) ethoxy] ethoxyl ethoxyl 2-methyl-2-propenoate (9CI) (CA

INDEX NAME)

CM 1

CRN 58985-96-9

```
 \begin{array}{c} {\rm H2C} \\ {\rm Me} - \underbrace{ \begin{array}{c} \\ \\ \end{array}}_{-} \underbrace{ \begin{array}{c} \\ \\ \end{array}}_{-} {\rm C} + {\rm I}_{2} - {
```

CM 2

CRN 2177-70-0 CMF C10 H10 O2

Pho\_U\_U\_Me

CM

CRN 88-12-0 CMF C6 H9 N O



RN 58986-00-8 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, 2-phenoxyethyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoste (9CI) (CA

propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM

CRN 58985-96-9 CMF C13 H22 O5

CM 2

CRN 10595-06-9 CMF C12 H14 O3 H2C O CH2 CH2 OPh

CM

CRN 88-12-0 CMF C6 H9 N O



58986-01-9 HCAPLUS

2-Propenoic acid, 2-methyl-, 4-methoxyphenyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-[2-propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA

INDEX NAME)

CM

CRN 58985-96-9 CMF C13 H22 O5

CM 2

CRN 10430-85-0 CMF C11 H12 O3



CM

3 CRN 88-12-0 CMF C6 H9 N O



58986-03-1 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, (4-methoxyphenyl)methyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM

CRN 58986-02-0 CMF C12 H14 O3



CM 2

CRN 58985-96-9 CMF C13 H22 O5

H2C 0 CH2 CH2 CH2 O CH2 CH2 O CH2 CH2 O CH2 CH2 O CH2 CH2 CH2 O CH2 CH2 CH2

CM

CRN 88-12-0 CMF C6 H9 N O



58986-04-2 HCAPLUS RN

propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA

INDEX NAME)

CM 1

62 Deleted: 3/6/2008

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CRN 58985-96-9
CMF C13 H22 O5
```

$$\begin{array}{c} ^{\rm H2C} \\ ^{\rm Me} = \begin{array}{c} 0 \\ ^{\rm H2C} = 0 \\ - \end{array} \\ \text{C} = 0 \\ - \end{array} \\ \text{C} \\ \text{H}_{2} = 0 \\ - \times \\ \text{C} \\ \text{H}_{2} = 0 \\ - \times \\ \text{C} \\ \text{H}_{2} = 0 \\ - \times \\ \text{C} \\ \text{H}_{2} = 0 \\ - \times \\ \text{C} \\ \text{H}_{3} = 0 \\ - \times \\ \text{C} \\ \text{C$$

CM 2

CRN 10475-46-4 CMF C14 H12 02



CM 3

CRN 88-12-0 CMF C6 H9 N O



RN 58986-05-3 HCAPLUS

NN 2-Propenoic acid, 2-methyl-, [1,1'-biphenyl]-4-yl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2-propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM

CRN 58985-96-9 CMF C13 H22 O5

CM 2

CRN 46904-74-9 CMF C16 H14 O2

CM

CRN 88-12-0 CMF C6 H9 N O



58986-07-5 HCAPLUS

2-Propenoic acid, 2-methyl-, 4-(phenylmethyl)phenyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58986-06-4 CMF C17 H16 O2

CH2-Ph



CM

CRN 58985-96-9 CMF C13 H22 O5

3 CM

CRN 88-12-0 CMF C6 H9 N O



58986-08-6 HCAPLUS

2-Propenoic acid, 2-methyl-, 4-(1,1-dimethylethyl)phenyl ester, polymer with 1-ethenyl-2-pyrrolidinone and 2-[2-[2-(2propenyloxy)ethoxy]ethoxy]ethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 58985-96-9 CMF C13 H22 O5

CM

CRN 13101-33-2 CMF C14 H18 O2

Н2С О

CM

CRN 88-12-0 CMF C6 H9 N O

ÇН<u>—</u> СН2

C08F; G02C

36-3 (Plastics Manufacture and Processing)

Section cross-reference(s): 63 58985-90-3 58985-91-4 58985-92-5 58985-93-6 58985-45-9 58985-97-€ 58985-96-1 58985-99-2

58986-08-5 58986-09-7

3/11/2008 10/588.210 65 Deleted: 3/6/2008

RL: USES (Uses) (hydrogels)

L35 ANSWER 26 OF 26 HCAPLUS COPYRIGHT 2008 ACS on STN ACCESSION NUMBER: 1955:27989 HCAPLUS Full-text

DOCUMENT NUMBER: ORIGINAL REFERENCE NO.: 49:5388e-1

Three dimensional polymerization of allyl ethers and mixed allyl ethers of methacrylic esters of glycols

AUTHOR(S): Berlin, A. A.; Dabagova, A. K.; Rodionova, E. F. SOURCE: Sbornik Statel Obshchel Khim, (1953), 2, 1560-5

LANGUAGE:

DOCUMENT TYPE: Unavailable Reaction of glycols with CH2:CMeCOCl or CH2:CHCH202CCl in the presence of pyridine at -10° gave 60-70% of the following derivs.: CH2:CHCH2CCH2CH2CH2C2CCMe:CH2, b2 69-70°, d2020 0.9706, nD20 1.4460; CH2:CHCH2CCH2CH2CCH2CCH2CCCMe:CH2, b2 85-6°, d2020 1.0270, nD20 1.4538; CH2:CHCH2O(CH2CH2O)2CH2CH2O2CCMe:CH2, b. undetd. owing to polymerization, d2020 1.0550, nD20 1.4587; CH2:CMeCO2CH2CH2O2CCH2CH:CH2, b2 106°, d2020 1.0820, nD20 1.4500; CH2; CMeCO2(CH2)20(CH2)2CO2CH2CH: CH2 b2 119-20°, d2020 1.1020, nD20 1.4560; CH2:CMeCO2(CH2CH2O)3COCH2CH:CH2, polymerizes on attempted distillation, d2020 1.1070, nD20 1.4585; (CH202CCH2CH:CH2)2, b2 127.5°, d2020 1.1210, nD20 1.4445; O(CH2CH2O2CCH2CH:CH2)2, b2 161°, d2020 1.1400, nD20 1.4515; CH2:CHCH2OCH2CH2O2CCH2CH:CH2, b2 85.5°, d2020 1.0384, nD20 1.4415. The intermediate allyl ethers were prepared from RCl or RBr and the corresponding Na derivative of the glycols; HOCH2CH2CH2CH2CH2CH2 (I), b. 159-60°, d2020 0.9526, nD20 1.4355; HO(CH2CH2O)2CH2CH:CH2, b2 98-101°, d2020 1.012, nD20 1.4440; HO(CH2CH2O)3CH2CH:CH2, b2 115-18°, d2020 1.0699, nD20 1.4530. Passage of ethylene oxide into CH2:CHCH2OH and 3% concentrated H2SO4 at 50-60° gave 50-5% yield of I. Polymerization of these esters were run in pure state and in 25% MeOH solns. The results, given graphically, show the following. The methacrylicallyl derivs, of the glycols and methacrylic-"carballylic" derivs. polymerize more rapidly than do "biscarballylic" or allyl "carballylic" derivs. Generally the increase of the distance between the functional groups of the above esters leads to increase rate of 3-dimensional polymerization; in "biscarballylic" esters this relationship is reversed. The principal factor affecting the rate of polymerization in MeOH is the steric factor which establishes the distance between the functional groups of the

56095-94-7, Ethanol, 2-[2-(allyloxy)ethoxy]-, methacrylate 38385-96-9, Ethanol, 2-[2-[2-(allyloxy)ethoxy]ethoxy]-, methacrylate (polymerization of)

58985-94-7 HCAPLUS

monomer.

2-Propenoic acid, 2-methyl-, 2-[2-(2-propenyloxy)ethoxylethyl ester (9CI) (CA INDEX NAME)

Me\_U\_U\_0\_CH2\_CH2\_O\_CH2\_CH2\_O\_CH2\_CH2\_CH

58985-96-9 HCAPLUS

2-Propenoic acid, 2-methyl-, 2-[2-[2-(2-

propenyloxy)ethoxylethoxylethyl ester (9CI) (CA INDEX NAME)

3/11/2LOS 10/588,210 66 Deleted: 3/6/2008

H2C 0 1e\_U\_U\_0\_0\_CH2\_CH2\_0\_CH2\_CH2\_0\_CH2\_CH2\_0\_CH2\_CH<u>=</u>CH2

CC 10 (Organic Chemiatry)

1 15075-50-0, Ethanol, 2-[2-(allyloxy)ethoxy]- 26150-05-0, Ethanol,
2-[2-[2-(allyloxy)ethoxy]ethoxy]- 44605-74-5, Carbonic acid, allyl
eater 63935-94-7, Ethanol, 2-[2-(allyloxy)ethoxy]-,
methacrylate 58945-94-9, Ethanol, 2-[2-[2(allyloxy)ethoxy]ethoxy]-methacrylate 58945-95-6-6,
3,6,9-Trioxadode-11-en-1-ol, methacrylate
(polymerization)

-8